

1 **Supplement:**

2 **Methods: Analytical Techniques**

3 Samples for the measurement of DMS/P/O concentrations and turnover rates were  
4 collected from sack hole brines (Thomas and Papadimitriou et al. 2003), ice covered  
5 seawater, surface slush, and melt ponds using a Rule Pump (02 bilge pump 1500GPH).  
6 Briefly, sack holes were drilled using an electric ice auger with a coring diameter of ~ 30  
7 cm. The sack holes were covered for 10-20 minutes, so that brine could percolate in from  
8 the surrounding ice walls while minimizing exposure full solar irradiance and inhibiting  
9 gas exchange. For depth profiles, the auger was lowered to several pre-determined  
10 depths below the ice surface. Snow and brine depth (cm) were measured from the sea ice  
11 surface with a meter stick. After complete drilling to the bottom of the sea ice, we  
12 lowered the pump to the seawater – ice interface to sample seawater immediately below  
13 the ice. In one case, we also collected surface seawater (~5 m depth) using the research  
14 vessel's underway seawater system. All samples were transferred into UV transparent  
15 (UVT), gas tight Welch Fluorocarbon 0.005" PFA bags, the headspace was removed, and  
16 the bags were clamped shut with Teflon closures. Samples were placed in opaque plastic  
17 bags and taken back to the shipboard laboratory for analysis within one hour of  
18 collection. Prior to analysis, samples were stored at 4°C.

19 Ancillary measurements were made on ice cores drilled within <1m from DMS/P/O  
20 depth profile cores. Chl *a* was measured fluorometrically [Holm-Hansen et al. 1965]  
21 using a Turner Fluorometer 10-AU (Turner Designs, Inc.) in triplicate subsamples from  
22 10cm ice core sections (7cm diameter), which were thawed in the dark in 2L of 0.2 mm  
23 filtered seawater (Arrigo et al., 2003). Samples were filtered onto 25mm GF/Fs, and the

24 filters were placed in 5mL of 90% acetone for 24 h extraction in the dark at  
25 4°C. Filtration volumes ranged from ~25ml to 250ml, depending on the biomass of  
26 samples. Chl *a* concentrations from depths  $\leq$  the sack hole brine depth were averaged for  
27 comparisons between Chl *a* levels with DMSPt because sackhole brine percolates from  
28 overlying and adjacent sea-ice brine channels. Ice temperature was measured  
29 immediately following core extraction on the ice and brine salinity was measured on  
30 board using a refractometer.

31 The analysis of DMS, DMSP<sub>t</sub>, DMSP<sub>d</sub> and DMSO<sub>t</sub> concentrations was conducted  
32 using a purge and trap gas chromatographic separation method, coupled to a capillary  
33 inlet mass spectrometer (PT-CIMS). Briefly, 30 ml subsamples were transferred from the  
34 Teflon bags into gas tight vials with Teflon-faced butyl seals. The vials were connected  
35 to a 16-position manifold valve (VICI Valco Instruments) and sequentially sparged for 5  
36 min at a rate of 450 ml min<sup>-1</sup> with UHP He to extract DMS onto a Carbopack-X trap held  
37 at room temperature. When the sparging was complete, the trap was rapidly heated to  
38 ~210°C to desorb DMS onto a fused capillary column. The effluent from the column was  
39 introduced, via capillary bypass inlet, into the electron impact ion source of a quadropole  
40 mass spectrometer (Hiden Analytical HAL 301), for detection using a secondary electron  
41 multiplier, with a voltage gain of 950 V. DMSP samples were analyzed after purging  
42 background DMS out of solution for ~10min with 450ml/min of UHP He (until no DMS  
43 remained) and a subsequent > 6 hour alkaline hydrolysis to DMS in 1M KOH. Samples  
44 for DMSP<sub>d</sub> determination were gently (~15 ml min<sup>-1</sup>) syringe filtered (Acrodisc 0.2µm)  
45 following the procedures described by [Kiene and Slezak, 2006] to minimize cell lysis.

46 DMSO<sub>t</sub> concentrations were measured using TiCl<sub>3</sub> reduction method to convert DMSO to  
47 DMS on samples stored at -20 °C for 6 weeks according to Kiene and Gerard [1994].

48 Unequal variance t-tests on the ranks (rather than Student's t-tests or Mann-Whitney U  
49 tests) were used to evaluate the difference between DMS/P/O concentrations in sea-ice  
50 and in ice-covered seawater because histograms of DMS/P/O data across the SIZ showed  
51 considerable skew and unequal variances between the groups [Ruxton, 2006]. We  
52 conducted four separate tests on the ranks of DMS, DMSP<sub>t</sub>, DMSP<sub>d</sub>, DMSO<sub>t</sub> across these  
53 two groups, and thus adopted an alpha of 0.001 for hypothesis testing. In addition to  
54 Pearson correlations, we used two multiple regressions between concentrations of  
55 DMS/P/O and ancillary parameters to probe for patterns in the observed variance: we  
56 regressed DMS against DMSO<sub>t</sub> and DMSP<sub>t</sub>, and we regressed DMSP<sub>t</sub> against Chl *a* and  
57 depth.

58 We quantified the rates of key production and consumption pathways in the DMS  
59 cycle using a multi-isotope tracer approach in sea ice brines (Fig S2). Sack hole brine  
60 samples were amended with deuterium-labeled dimethyl <sup>2</sup>H<sub>3</sub>-DMS (CDN Isotopes, 99%  
61 purity), <sup>2</sup>H<sub>6</sub>-DMSP<sub>d</sub> (synthesized from <sup>2</sup>H<sub>3</sub>-DMS, CDN Isotopes, 99% purity) and Fluka  
62 3-bromopropionic acid using the method of Challenger and Simpson (1948) (*Dacey and*  
63 *Stefels*, 2005), and <sup>13</sup>C<sub>2</sub>-DMSO<sub>d</sub> (Icon Isotopes, 97% purity). Tracers were added to  
64 obtain final concentrations ~10% of natural concentrations. All tracers were added  
65 simultaneously into duplicate or triplicate experimental bags, which were incubated for 3-  
66 4 hours outside in coolers packed with ice to maintain in situ temperatures (-1°C). We  
67 used various layers of neutral density screening in addition to snow cover (mimicking *in*

68 *situ* thickness) to reduce ambient irradiance in the coolers to levels similar to those at  
69 various depths within the ice.

70 We replicated *in situ* temperatures and irradiance levels as closely as possible during  
71 these incubations to examine natural rates of DMS cycling, although it is possible that  
72 certain phytoplankton and larger organisms that adhere onto the sidewalls of sea ice brine  
73 channels may have been under sampled during our incubations (Thomas and  
74 Papadimitriou et al. 2003). Nevertheless, we deemed sack hole brine sample collection  
75 for these incubations our best option for these incubations. Slowly melting of sea ice  
76 (Arrigo et al., 2003), would likely alter the rates of DMS consumption and production by  
77 influencing microbial dynamics, and promoting gas exchange. Similarly, sample  
78 collection based on ice crushing (*Tison et al.*, 2010), perturbs the ecosystems maintained  
79 within brine channels.

80 During experiments, subsamples were extracted from each bag with 60 ml syringes  
81 every ~30-40 minutes, filtered with sterile 0.45 mm syringe filters into 50 ml amber  
82 serum bottles, capped with Teflon-faced butyl septa and loaded onto a multiport valve for  
83 analysis as described above. Between uses, UVT incubation bags were rinsed once with  
84 ~10% HCL, fresh MQ, and twice with each sample to avoid contamination. In addition,  
85 serum bottles used for analysis were soaked in ~5% HCl for ~24 hrs, rinsed with fresh  
86 MQ, and dried in a muffled heater at 125°C for ~1hr prior to use [*Kiene and Gerard*,  
87 1994].

88 DMS produced from DMSP and DMSO was analyzed as described above.  
89 Concentrations of isotopically-labeled species ( $^{13}\text{C}_2\text{-DMS}$ ,  $^2\text{H}_3\text{-DMS}$ , and  $^2\text{H}_6\text{-DMS}$ )  
90 were calculated by integrating chromatogram signals of different isotopic DMS species

91 measured by peak jumping the mass spectrometer between  $m/z$  62, 64, 65 and 68. Final  
92 concentrations were calculated from standard curves using known concentrations of both  
93 unlabeled and labeled DMS with a working background detection of 0.1nM for 30 ml  
94 samples. Integrated peak areas were converted to DMS concentrations using standard  
95 additions of DMS prepared in deep seawater (>1500m), with intermediate dilutions  
96 prepared in milli-Q water. Tracer production and consumption rates were quantified with  
97 linear regressions of averaged concentrations for 2-3 replicates (one for each bag that was  
98 incubated) over 4-5 time points and converted to  $\text{nM d}^{-1}$ . Rates that were not statistically  
99 different from 0 (with  $p>0.1$ ) were considered below the detection limit. To scale the  
100 rates of tracer consumption and production to *in situ* values, the calculated rates were  
101 divided by the concentration of added tracers (yielding first order rate constants,  $\text{d}^{-1}$ ) and  
102 multiplied by the concentration of the natural  $\text{DMS}_d$  / $\text{DMSP}_d$  / $\text{DMSO}_d$  pools respectively.  
103 Concentrations of unlabeled  $\text{DMSO}_d$  and DMS were measured at the start of each  
104 incubation, and  $\text{DMSP}_d$  concentrations were inferred from our depth profile  
105 measurements. Net DMSO turnover rates were calculated from the difference in  $^{13}\text{C}_2$ -  
106 DMSO concentrations at the beginning and end of the incubations, and converted into  
107  $\text{nM/day}$ . DMSO-DMS yields were then estimated as the fraction of DMSO converted to  
108 DMS. The calculated DMSO-DMS yield are subject to significant uncertainty due error  
109 propagation and analytical variability. The rates of change of deuterated and  $^{13}\text{C}$ -labeled  
110 stable isotope tracers are assumed to be representative of natural DMS cycling because  
111 isotope discrimination between species of DMS, favoring lighter isotopes, appears  
112 minimal (<10%; Asher unpublished data).

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114 **References**

- 115 Arrigo, K.R., R.H. Dale, R.B. Dunbar, A.R. Leventer, M.P. Lizotte (2003), Physical  
116 control of chlorophyll a, POC, and TPN distributions in the pack ice of the Ross Sea,  
117 Antarctica. *J. Geophys. Res.*, 108 (C10) 3316-3330. doi: 10.1029/2001JC001138
- 118 Challenger F, Simpson MI (1948), Studies on biological methylation. A precursor of the  
119 dimethyl sulphide evolved by *Polysiphonia fastigiata*. Dimethyl-2-carboxyethyl  
120 sulphonium hydroxide and its salts. *J. Chem. Soc.* 1948,1591–1597.
- 121 Dacey, J.W.H., and J. Stefels (2005), Deuterated tracers for dynamics of the DMS system  
122 in marine waters ASLO, Santiago, Spain.
- 123 Holm-Hansen, O., C. J. Lorenzen, R. W Holmes, J. D. H. Strickland (1965), Fluorimetric  
124 Determination of Chlorophyll, *ICES J. Mar. Sci.*, 30(1), 3-15.
- 125 Hatton, A. D., et al. (2004), The role of dimethylsulphoxide in the marine biogeochemical  
126 cycle of dimethylsulphide, in *Oceanogr. Mar. Bio. Ann. Rev.*, Vol 42, edited, pp. 29-55.
- 127 Kiene, R. P., and G. Gerard (1994), Determination of trace levels of dimethylsulfoxide  
128 (DMSO) in seawater and rainwater, *Mar. Chem.*, 47(1), 1-12.
- 129 Kiene, R. P., and D. Slezak (2006), Low dissolved DMSP concentrations in seawater  
130 revealed by small-volume gravity filtration and dialysis sampling, *Limnol. Oceanogr.*  
131 *Meth.*, 4, 80-95.
- 132 Ruxton, G.D. (2006), The unequal variance t-test is an underused alternative to the  
133 student's t-test and the Mann-Whitney U-test, *Behav. Ecol.*, 17(4), 688-690.
- 134 Stefels, J. et al. (2007), Environmental constraints on the production and removal of the  
135 climatically active gas dimethylsulphide (DMS) and implications for ecosystem  
136 modeling, *Biogeochem.*, 83(1-3), 245-275.

137 Thomas, D. N., and S. Papadimitriou (2003), Biogeochemistry of sea ice, in *Sea Ice: An*  
138 *Introduction to Its Physics, Biology, Chemistry and Geology*, edited by D. N. Thomas and  
139 G. S. Dieckmann, pp. 267–302, Blackwell Science, Oxford, UK.

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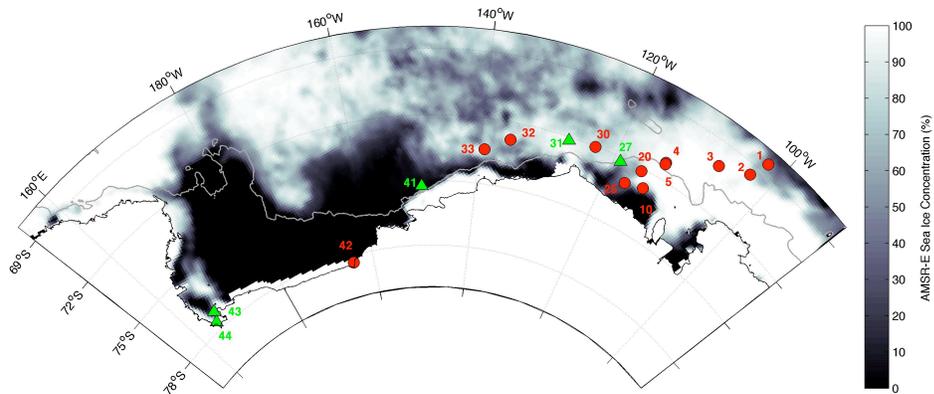
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## 148 **Supplementary Figures**

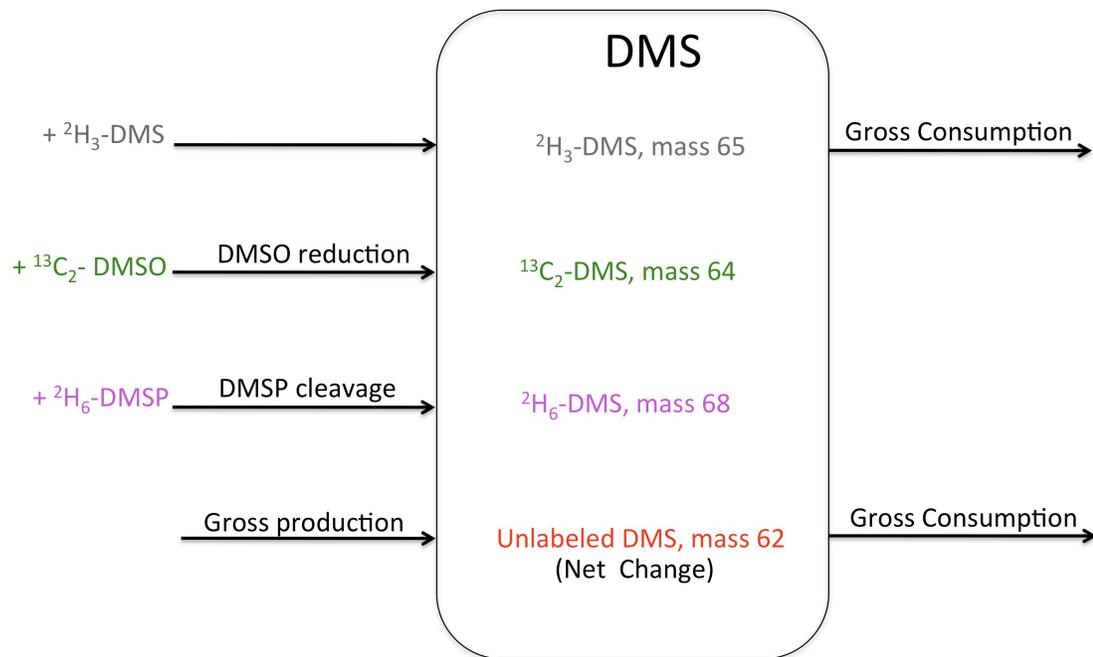
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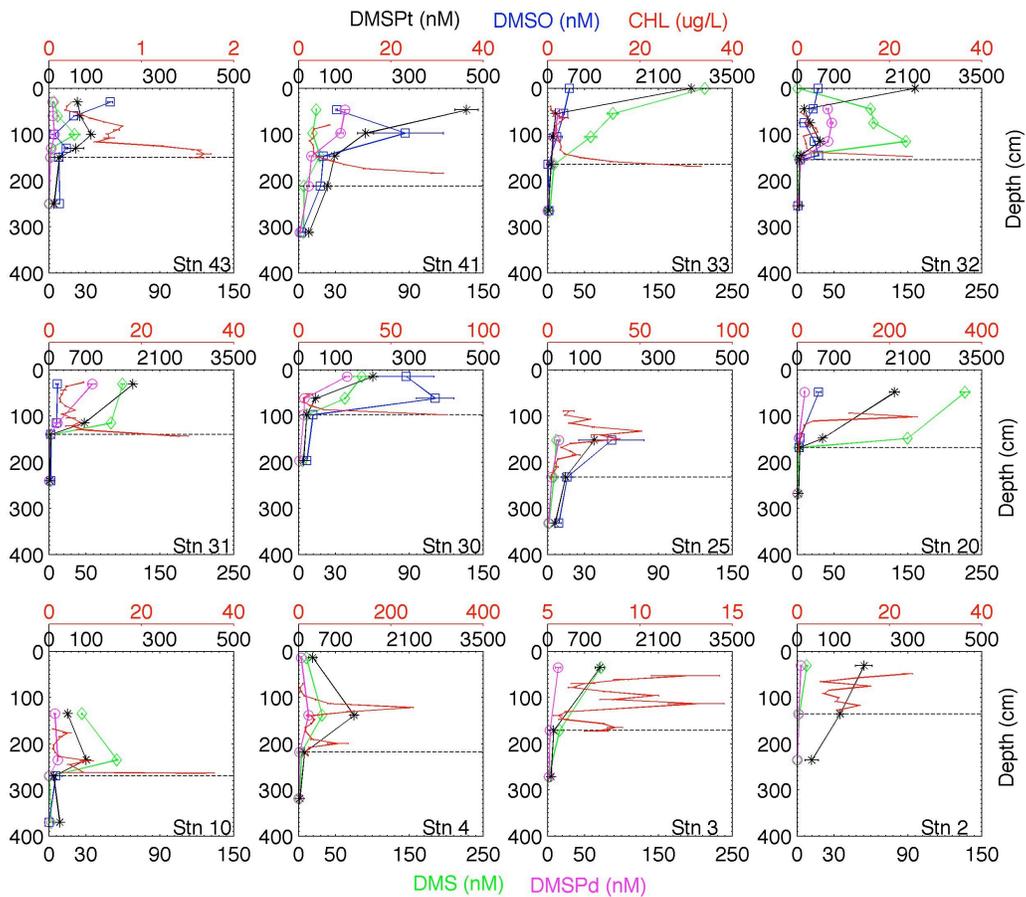
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151 **Figure S1.** Map of sampling stations across an east to west transect through the SIZ of  
152 the Amundsen Sea and Ross Sea between December 16, 2010 and January 10, 2011.

153 Station numbers appear next to station locations. Green triangles denote stations where  
154 isotope tracer experiments were conducted in addition to DMS/P/O measurements. Red  
155 circles denote stations where only concentration measurements were made. The  
156 background black and white color scale shows the mean sea ice concentrations during the  
157 period of our survey derived from the AMSR-E satellite [Cavelieri et al. 2004].



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159 Fig. S2. Schematic diagram of the DMS tracer method. Tracers, shown in gray, green,  
160 and purple are added to the ambient DMS,  $\text{DMSO}_d$  and  $\text{DMSP}_d$  pools, respectively, and  
161 detected in the in DMS pool. Labeled black arrows denote measured production and  
162 consumption processes affecting each isotopic mass of DMS.  
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166 Fig. S3. Depth profiles of methylated S compounds at 12 sea ice stations with  
 167 chlorophyll *a* concentrations (in red) for comparison. Blue squares and lines denote  
 168  $\text{DMSO}_t$  concentrations, while black stars represent  $\text{DMSP}_t$  (plotted along the top *x*-axis).  
 169 Green diamonds and lines represent DMS and magenta circles represent  $\text{DMSP}_d$ , which  
 170 are plotted along the bottom *x*-axis. Dashed horizontal lines on the plots show the sea-  
 171 ice/seawater interface, and station numbers are marked in the bottom right of each  
 172 subplot. Note that all subplots do not have identical *x*-axes.

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