

1 **Geochemistry of surface sediments from the fjords of Northern Chilean Patagonia (44-**
2 **47°S): Spatial variability and implications for paleoclimate reconstructions**

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16 **Abstract (287 words)**

17 The Patagonian fjords have a clear potential to provide high-resolution sedimentary and
18 geochemical records of past climate and environmental change in the Southern Andes. To
19 improve our ability to interpret these proxy records, we investigated the processes that control
20 fjord sediment inorganic geochemistry through a geochemical, mineralogical and
21 sedimentological analysis of surface sediment samples from the fjords of Northern Chilean
22 Patagonia. A simple terrestrial index based on measurements of salinity and fraction of
23 terrestrial carbon was used to estimate the terrestrial input/river discharge at each site. Our
24 results demonstrate that, under the cold climate conditions of Patagonia, chemical weathering
25 is weak and the inorganic geochemical composition of the fjord sediments is primarily
26 controlled by hydrodynamic mineralogical sorting, i.e., the intensity of river discharge. Our
27 results suggest that the distribution of Fe, Ti and Zr in surface sediments is controlled by their
28 association with heavy and/or coarse minerals, whereas Al is independent of hydrodynamic
29 processes. The elemental ratios Fe/Al, Ti/Al and Zr/Al are therefore well suited for estimating
30 changes in the energy of terrestrial sediment supply into the fjords through time. Zr/Al is
31 particularly sensitive in proximal environments, while Fe/Al is most useful in the outer fjords
32 and on the continental margin. In the most proximal environments, however, Fe/Al is
33 inversely related to hydrodynamic conditions. Caution should therefore be exercised when
34 interpreting Fe/Al ratios in terms of past river discharge. The application of these proxies to
35 long sediment cores from Quitralco fjord and Golfo Elefantes validates our interpretations.
36 Our results also emphasize the need to measure Al-based elemental ratios at high precision,
37 which can be achieved using simultaneous acquisition ICP-AES technology. This study
38 therefore constitutes a strong basis for the interpretation of sedimentary records from the
39 Chilean Fjords.

40 **1. INTRODUCTION**

41 Understanding the causes of past climate variability requires an array of high-resolution
42 paleoclimate records with geographical and chronological resolutions adequate to analyze
43 past patterns of climate dynamics. In this respect, the mid- and high-latitudes of the Southern
44 Hemisphere have been relatively understudied, although they play a critical role in our
45 understanding of Earth's climate variability. At these latitudes the climate is dominated by the
46 latitudinal position and strength of the Southern Westerly Winds (SWW; Garreaud et al.,
47 2009). Since Southern South America (SSA) is the only continuous land mass in the Southern
48 Hemisphere that intersects the entire westerly wind belt, it constitutes a key region for
49 paleoclimate reconstructions (e.g., Toggweiler, 2009).

50 Reconstructions of terrestrial climate in SSA are mostly limited to a few lake sedimentary
51 records from Argentina (e.g., Gilli et al., 2005; Mayr et al., 2007), the Chilean Lake District
52 (e.g., Bertrand et al., 2008; Moreno et al., 2010), and the southern tip of Chile (e.g., Moy et
53 al., 2008; Waldmann et al., 2010; Moreno et al., 2010), to discontinuous records of glacial
54 activity (e.g., Kaplan et al., 2008; Moreno et al., 2009) and to a limited number of Late
55 Holocene tree-ring records from Northern Patagonia (e.g., Lara and Villalba, 1993; Villalba et
56 al., 1997). In addition, current paleoceanographic records are limited to a very few sites from
57 the southeastern Pacific, mainly from latitudes around or North of 40°S (e.g., Lamy et al.,
58 2004, Kaiser et al., 2005, 2008; Mohtadi et al., 2007; Muratli et al., 2010). Very few papers
59 present paleoenvironmental reconstructions based on fjord and near-shore sediments from
60 Chilean Patagonia (Sepúlveda et al., 2009; Lamy et al., 2010; Siani et al., 2010). Our current
61 understanding of past changes in SSA climate remains therefore limited and additional high-
62 resolution and continuous records are critically needed, especially from the "Roaring Forties"
63 latitudes (e.g., Villalba et al., 2009, Moreno et al., 2010).

64 Motivated by this lack of data, several research teams recently set out to study the
65 sedimentary record of the Chilean Fjords and the adjacent continental margin, and a large
66 number of sediment cores were successfully retrieved during oceanographic cruises, such as
67 the Cimar-7-Fiordo (e.g., Sepúlveda et al., 2009), the NB Palmer 0505 (e.g., Boyd et al, 2008),
68 the R/V Mirai BEAGLE-2003 and MR08-06, and the R/V Marion Dufresne PACHIDERME.
69 Sediments from the Chilean fjords are particularly promising because they present high
70 accumulation rates (Salamanca and Jara, 2003; Fernandez et al., in press) and can record
71 changes in river discharge (e.g., Sepúlveda et al., 2009), which is linked to precipitation and
72 glacier melting in the Andes (Dávila et al., 2002).

73 With this in mind, the goal of the current study is to investigate the processes that control
74 sediment geochemistry in the Chilean fjords, with a particular focus on the lithophile and
75 mostly immobile elements Al, Fe, Ti, and Zr (Mc Lennan et al., 2003), since these elements
76 are typically associated to the lithogenic fraction of the sediment and are frequently used as
77 indicators of terrestrial supply in sediment core studies (e.g., Haug et al., 2001, Lamy et al.,
78 2004). After assessing the ability of ICP-AES technology to measure inorganic elemental
79 ratios at high precision, we present geochemical, mineralogical and sedimentological data
80 obtained on a series of surface sediment samples from the fjords of Northern Chilean
81 Patagonia. These multi-proxy analyses, which include the parameters that are the most
82 frequently measured on sediment cores, are then used to discuss (1) the natural parameters
83 that control the bulk composition and the inorganic geochemistry of the sediment, and (2) the
84 best proxies for reconstructing past changes in the energy of river sediment discharge. We
85 deliberately selected samples from fjords in different geomorphological and glaciological
86 settings since, to be useful for paleoclimate reconstructions, the proposed proxies have to be
87 (1) applicable to sediment cores from any Northern Patagonian fjord, and (2) largely
88 independent of accumulation rates, watershed size, and variations in the nature of the bedrock

89 and soil cover. Our research builds on bulk organic geochemical results previously obtained
90 on surface sediments from the Patagonian fjords (Silva and Prego, 2002; Sepúlveda et al.,
91 2011; Silva et al., 2011), and on multi-proxy studies of surface sediment samples from the
92 Chilean continental margin between 25 and 43°S (Lamy et al., 1998; Hebbeln et al., 2000;
93 Klump et al., 2000; Romero et al., 2001).

94

95

2. REGIONAL SETTING

96 The geology of Northern Chilean Patagonia is dominated by the North Patagonian Batholith,
97 which forms the core of the southern Andes and is roughly parallel to the coast (Fig. 1). It is
98 composed of Cenozoic and Mesozoic granitoids, mainly in the form of hornblende-biotite
99 granodiorites and tonalities (Pankhurst et al., 1999; Parada et al., 2007). The batholith is
100 flanked by Mesozoic metamorphic rocks to the West, which form the western side of the
101 Chonos Archipelago and Taitao Peninsula (Fig. 1), and by Mesozoic volcanic rocks to the
102 East (Parada et al., 2007; Sernageomin, 2003). One of the most striking structural features in
103 the region is the Liquiñe-Ofqui fault system, which is responsible for the first-order
104 morphology of the fjords (Glasser and Ghiglione, 2009), and controls the location of the
105 regional volcanoes (Stern et al., 2007). Five of the thirteen Quaternary volcanoes that
106 compose the southern segment of the southern volcanic zone (SSVZ, 42-46°S) are located in
107 our study region (volcanoes Melimoyu, Mentolat, Cay, Macá and Hudson; Fig. 1). Except for
108 Cay volcano, all the SSVZ volcanoes have erupted during the Holocene, Hudson being by far
109 the most active (Stern et al., 2007). These volcanoes are mainly composed of lavas and
110 pyroclasts of basaltic to dacitic composition (Naranjo and Stern, 1998; D’Orazio et al., 2003).
111 The regional soil cover is dominated by andosols, i.e., soils developed on volcanic deposits
112 (Chile, 2003; Gut, 2008). The southern part of the study region is covered by the Northern

113 Patagonian Ice Field (NPI), a 4200 km² ice cover that is composed of 70 glaciers larger than
114 0.5 km² (Fig. 1; Rivera et al., 2007).

115 The morphology of the fjords is complex, with narrow channels and numerous islands
116 separating the mainland from the adjacent Pacific Ocean (Fig. 1). Bathymetric surveys have
117 demonstrated the presence of deep valleys reaching ~600 m at the mouth of the main fjords,
118 and numerous shallow sills of morainic origin (Araya-Vergara, 1997; Rodrigo, 2008). The
119 presence of these sills limits the exchange of bottom waters with the Pacific Ocean, favors
120 high sedimentation rates, and promotes the preservation of organic matter, although anoxic
121 basins have never been observed (Sepúlveda et al., 2005; Silva and Guzman, 2006; Sievers
122 and Silva, 2008). Water circulation in the fjords is generally described as a two-layer system,
123 with a surface estuarine water mass (Chilean Fjord Water, CFW) flowing out of the fjords
124 between 0 and ~30 m depth, and a deeper and more saline subantarctic water mass flowing in
125 opposite direction between ~30 m and ~150 m depth (Silva and Guzman, 2006; Sievers and
126 Silva, 2008). Where shallow sills do not restrict water circulation, a third deeper and warmer
127 layer composed of modified equatorial subsurface water also occurs (Silva and Guzman,
128 2006; Sievers and Silva, 2008). The extension and depth of the CFW depends on the amount
129 of freshwater supplied by rivers, glaciers, coastal runoff and direct precipitation, and its
130 salinity reflects the distance from the fresh water sources (Dávila et al. 2002; Sievers and
131 Silva, 2008). The CFW also contributes to the freshening of the northward-flowing Chilean
132 Coastal Current (Strub et al., 1998).

133 The climate of Northern Chilean Patagonia is strongly oceanic, with high precipitation
134 originating from the combination of the strong SWW with the rough topography of the
135 Andes. Precipitation shows a low seasonality and a strong West-to-East gradient, ranging
136 from ~3000 mm/yr on the western side of the Andes to less than 600 mm/yr at the border with
137 Argentina (Miller, 1976; Aravena and Luckman, 2009). The high precipitation in the area is

138 responsible for a high input of freshwater and terrigenous material to the fjords by river
139 discharge and terrestrial runoff. The fragmentary hydrographic data (Dirección General de
140 Aguas, Chile) show annual average river discharge of 515 and 256 m³.s⁻¹ for Rio Aysén and
141 Rio Cisnes, respectively (Calvete and Sobarzo, 2011). Most of the fjords also receive
142 freshwater and terrestrial material from secondary, smaller tributaries located along their
143 profiles. The average air temperature is 8-9°C, with the highest values occurring in January
144 (13°C) and the lowest in July (4-6°C).

145

146

3. MATERIAL AND METHODS

147 3.1 Sampling and sample preparation

148 Surface sediment samples were obtained with a box-corer during the CIMAR (Cruceros de
149 Investigación Marina) 7 Fiordos expedition (CF7) in November 2001, aboard the R/V AGOR
150 Vidal Gormaz. Samples were collected at 14 stations between 43.7°S and 46.5°S (Fig. 1,
151 Table 1) at depths varying between 52 and 582 m. The sampling stations are roughly located
152 along a terrestrial-marine gradient, from the head of the inner fjords to the break of the
153 continental shelf (Fig. 1). The box-cores were sub-sampled on board, using 7-cm diameter
154 PVC tubes, and the samples used in this study consist of a slice of the 0-1 cm depth interval.
155 Samples were stored in plastic bags and frozen at -20 °C until laboratory analyses. Before
156 analysis, all samples were freeze-dried and gently ground and homogenized in an agate
157 mortar. According to ²¹⁰Pb profiles, all surface sediment samples represent modern sediments
158 (Rojas, 2002; Salamanca and Jara, 2003; Rebolledo et al., 2005; Sepúlveda et al. 2005;
159 Rebolledo, 2007).

160 To represent the terrestrial end-member of the sediment, this study uses five river sediment
161 samples that were collected ~100 m upstream of river mouths during the CF7 cruise, and 7
162 samples from 4 soil profiles that were collected in the fjord watersheds in 2007 and 2008 (Fig.

163 1; Table 1). In the laboratory, the river sediment samples were freeze-dried, and the soil
164 samples were oven-dried at 50°C. These samples were subsequently dry-sieved at 125 µm to
165 discard the coarse particles that are not representative of the sediment fraction that reaches the
166 fjords. They were homogenized in an agate mortar before analysis. Data from the literature
167 (Lamy et al., 1998; Hebbeln et al., 2000; Klump et al., 2000) were used to represent the distal
168 (open ocean) end-member of the sediment.

169 Finally, samples from sediment cores JPC14 (n=44) and PC29A (n=51), which were collected
170 in Quitralco fjord and Golfo Elefantes, respectively (Fig. 1), were analyzed to test the
171 applicability of our results on long cores. These samples were also freeze-dried and gently
172 ground and homogenized in an agate mortar before geochemical analysis. Sediment core
173 JPC14 (46.449°S–73.798°W) is a 15m long jumbo piston core that was collected at 129 m
174 depth in the central basin of Golfo Elefantes. The core was taken during cruise NBP05-05 in
175 2005, on board the RVIB N.B. Palmer. It is composed of a 3m thick sand unit, surrounded by
176 fine-grained sediment. It covers the last 5400 years and essentially contains sediment
177 delivered by a proglacial river system (Bertrand et al., 2011a). Core PC29A is a 208 cm long
178 piston sediment core collected at station 29A (45.756°S–73.467°W; 112m depth) during
179 cruise CF7 (see above). It is entirely composed of fine-grained (silt) sediment and it
180 represents the last 1400 years (Bertrand et al., 2011b). It was collected in front of Rio Pelu, a
181 small river that drains a 128 km² unglaciated watershed (Ghazoui et al., 2011).

182

183 **3.2 Inorganic geochemistry**

184 Samples were prepared using the Li Metaborate fusion technique following Murray et al.
185 (2000), which is preferred over HF digestion because it is the only technique that allows the
186 complete dissolution of sediment samples containing refractory minerals such as zircon
187 (Sholkovitz, 1990; Murray et al., 2000; Huang et al., 2007). Sample preparation consisted in

188 mixing 200±1.0 mg of ultrapure Li-metaborate (SCP Science) in 3ml Pt:Au (95:5) crucibles,
189 with 50±0.5 mg of sediment. Ten µl of 25% LiBr were then added to the mixture and the
190 crucibles were placed in a muffle furnace for 12 minutes at 1050°C. The newly formed glass
191 bead was then allowed to cool down for 2-3 min, detached from the crucible, and poured into
192 a Teflon beaker containing a swirling 25 ml solution of 5% HNO₃. Complete dissolution
193 occurred within ~30 min. The solution was then filtered through a 0.45 µm PVDF Millipore
194 filter and diluted in 5% HNO₃ to obtain a 4000 x final dilution. The exact dilution factor was
195 calculated from the precise weight of sediment used for fusion.

196 Thirteen elements (Electronic Annex EA-1) were measured on a JY Ultima C ICP-AES
197 equipped with a mono- and a poly-chromator that were used in parallel. The analytical
198 conditions (Nebulizer type and flow, pump speed, argon pressure, and gas humidifier) of
199 Murray et al. (2000) were strictly applied. The thirteen elements were analyzed on both the
200 mono- and poly-chromator, except for P, for which no wavelength was available on the
201 polychromator. Measurements were made in triplicates and the measured concentrations were
202 corrected for instrumental drift using the measured concentrations of a matrix-matched
203 standard solution ran after every sample. Accuracy and analytical precision (EA-1) were
204 calculated from the analysis of ten individually-prepared sub-samples of reference sediment
205 PACS-2. PACS-2 was selected because its geochemical composition matches the average
206 composition of the fjord sediment samples. The results are presented in Table 2, EA-2 and
207 EA-3.

208

209 **3.3 Bulk organic geochemistry**

210 Approximately 50 mg of ground sediment was weighed in tin capsules and treated with 1N
211 sulphurous acid to remove eventual carbonates (Verardo et al., 1990). Total Organic
212 Carbon (TOC), Total Nitrogen (TN) and stable isotope ratios of carbon ($\delta^{13}\text{C}$) and nitrogen

213 ($\delta^{15}\text{N}$), were measured at the UC Davis Stable Isotope Facility by continuous flow isotope
214 ratio mass spectrometry (CF-IRMS; 20-20 SERCON mass spectrometer) after sample
215 combustion to CO_2 and N_2 at 1000°C in an on-line elemental analyzer (PDZ Europa
216 ANCA-GSL). The working standards, which are periodically calibrated against
217 international isotope standards (IAEA N1, N3; IAEA CH7, NBS22), were a mixture of
218 ammonium sulfate and sucrose with $\delta^{15}\text{N}$ vs Air = 1.33 ‰ and $\delta^{13}\text{C}$ vs PDB = -24.44 ‰.
219 The precision, calculated by replicate analysis of the internal standard, is 0.03 ‰ for $\delta^{13}\text{C}$
220 and 0.08 ‰ for $\delta^{15}\text{N}$. Some of the bulk organic geochemical results used in this study were
221 previously reported in Sepúlveda et al. (2011).

222

223 **3.4 Biogenic opal**

224 Biogenic silica (bio-Si) was analyzed according to Carter and Colman (1994) and Mortlock
225 and Froelich (1989). Samples were extracted with NaOH after removal of organic matter and
226 carbonate with 10% H_2O_2 and 1N HCl, respectively. They were subsequently diluted in 5%
227 nitric acid and analyzed in triplicate for Si, Na and Fe, on a JY Ultima-C ICP-AES. Al was
228 measured in triplicate by flame atomic absorption spectrometry (FAAS) on a Varian
229 SpectrAA 220. Measured Si concentrations were corrected for detrital Si using the measured
230 Al concentrations: $\text{bio-Si} = \text{measured Si} - 2 \times \text{Al}$. The 2:1 ratio accounts for Si leached from
231 volcanic glasses and clay minerals, and this value is similar to the Si:Al ratio measured on soil
232 samples prepared with the same technique. This correction assumes that all Al originates from
233 the dissolution of detrital particles. The analytical precision, determined from 5 entirely
234 separate analyses of a sediment sample from site BC29A, was 1.15 % for Si and 3.23 % for
235 Al. The precision for bio-Si, i.e. after correction for detrital Si, reached 0.46 wt. %, which
236 indicates that the correction procedure also corrects for uncertainties associated with sample
237 preparation. Biogenic opal (bio-opal, $\text{SiO}_2 \cdot n\text{H}_2\text{O}$, wt. %) was obtained by multiplying the

238 bio-Si values by 2.4 (Mortlock and Froelich, 1989). Lithogenic silica (litho-Si, wt. %) was
239 calculated by difference (Total Si [see 3.2] – bio-Si).

240

241 **3.5 Carbonate content**

242 The weight percentage of total inorganic carbon (TIC) in bulk sediment samples was
243 determined using an UIC CM5014 coulometer equipped with a CM5130 acidification
244 module. For each sample, 50-60 mg of sediment was precisely weighed in a 4ml glass vial
245 and treated with 1.5 ml 1N H₃PO₄ to liberate CO₂. The percentage of carbonate was
246 calculated from the TIC data using the following equation: CaCO₃ (wt%) = TIC (wt%) x 8.33,
247 assuming that 100% of the measured CO₂ is derived from dissolution of calcium carbonate.
248 The analytical precision, determined from 7 entirely separate analyses of a sediment sample
249 from site BC29A, was 0.04% CaCO₃.

250

251 **3.6 Grain size**

252 Grain size was measured on the terrigenous fraction of the sediment using a Coulter LS200
253 laser grain size analyzer. The terrigenous fraction was isolated by treating the samples with
254 boiling H₂O₂, HCl and NaOH, to remove organic matter, carbonates and biogenic silica,
255 respectively. Prior to analysis, samples were boiled with 300 mg of sodium pyrophosphate
256 (Na₄P₂O₇ · 10H₂O) to ensure complete disaggregation of the particles. The grain size
257 distribution of the samples was measured during 90 seconds and the arithmetic mean was
258 calculated from the 92 size classes.

259

260 **3.7 Bulk and clay mineralogy**

261 Bulk and clay mineralogy was analyzed by X-ray diffraction (XRD) on a Bruker D8-Advance
262 diffractometer with CuK α radiation. A first aliquot was separated and mounted as unoriented

263 powder by the back-side method (Brindley and Brown, 1980), and subsequently scanned by
264 XRD between 2° and $45^\circ 2\theta$. Peak intensities were used to quantify (± 5 wt. %) the mineral
265 proportions, following Cook et al. (1975). Clay minerals were not identified or quantified on
266 the bulk diffractograms. Although halite was detected in all the surface sediment samples, it
267 was not quantified because it derives from interstitial water salts that precipitated during
268 freeze-drying. Clay mineralogy was analyzed on the decarbonated sediment fraction $< 2 \mu\text{m}$.
269 Sample preparation consisted in wet-sieving at $63 \mu\text{m}$, decarbonation with 0.01 N HCl , and
270 removal of organic matter with H_2O_2 . Samples were then rinsed twice with DI water and the $<$
271 $2 \mu\text{m}$ fraction was separated from the aqueous suspension using the pipette method (1 cm
272 after 50 min, according to Stokes settling law with $d=2.65$). Oriented mounts were prepared
273 by the “glass-slide method” (Moore and Reynolds, 1989) and subsequently scanned on the
274 diffractometer between 2° and $30^\circ 2\theta$ after air drying at room temperature, between 2° and
275 $30^\circ 2\theta$ after solvation with ethylene-glycol for 24h, and between 2° and $15^\circ 2\theta$ after oven-
276 heating at 500°C during 4h. In addition, slides solvated with Ethylene-glycol were scanned at
277 slow speed between 23 and $27^\circ 2\theta$. Diffractograms were interpreted according to Petshick et
278 al. (1996) and the mineral proportions were quantified (wt. %) in MacDiff v 4.2.5, following
279 the peak area method of Biscaye (1965). Five clay minerals were identified: smectite, illite,
280 chlorite, kaolinite and vermiculite.

281

282 **3.8 Magnetic susceptibility**

283 Magnetic susceptibility (MS) is one of the most frequently measured parameters in sediment
284 cores, owing to the speed and cost effectiveness of the method. Bulk MS is usually used as an
285 indicator of the concentration of allochthonous mineral matter in sediments (e.g., Sandgren
286 and Snowball, 2001). It is a measure of the net contribution of ferromagnetic (magnetite,
287 hematite, etc) and paramagnetic (olivine, pyroxene, amphibole, etc) minerals, with the

288 contribution of the latter being more important when ferromagnetic minerals occur in very
289 low concentrations (Houdra and Kahan, 1991; Sandgren and Snowball, 2001). Volume
290 magnetic susceptibility was measured with a Bartington MS2G single-frequency (1.3 kHz)
291 sensor, connected to a Bartington MS3 meter. Sediment samples were gently packed into 1 ml
292 plastic vials and were analyzed in duplicate.

293

294 **3.9 Statistical analyses**

295 Statistical analyses, including Pearson correlation coefficients, p-values (two-tailed test of
296 significance), Shapiro-Wilk normality tests, Principal Component Analysis (PCA) and
297 Redundancy Analysis (RDA) were conducted with XLSTAT v. 2010.3 (EA-4 and EA-5). The
298 soil and river sediment samples were not included in statistical analyses to prevent any bias
299 by entirely terrestrial samples. Except where indicated, correlations with $p < 0.05$ were
300 considered significant. The PCA dataset consisted of the 52 variables that are presented in
301 EA-2. Statistical analyses were also conducted on a restricted dataset including samples from
302 Puyhuapi and Jacaf fjords only (i.e., sites 33, 35, 36, 39, 40 and 42; see Fig. 1) to assess the
303 influence of regional variability in lithology on the geochemical results (EA-5). The results of
304 the RDA were essentially similar to those of the PCA.

305

306 **4. RESULTS AND INTERPRETATION**

307 **4.1 Analytical precision and elemental ratios**

308 The analytical precision obtained on the elemental concentrations with the polychromator is
309 always better than or similar to the precision obtained with the monochromator (EA-1). The
310 limits of detection (LODs), on the other hand, are generally, but not systematically, better for
311 the monochromator lines (EA-1). For fjord sediment samples, and for sediment and
312 sedimentary rocks in general, the concentrations of the 13 elements of interest are far above

313 (>25x) the LODs of the poly- and mono-chromator lines (EA-1). The only two exceptions are
314 K (very high LOD on the polychromator due to the broad peak associated to the high
315 wavelength of the K line) and Zr (concentration in most sediment samples is only 10-20 x
316 LOD). Data obtained with the polychromator lines were therefore selected for Al, Ba, Ca, Fe,
317 Mg, Mn, Na, Si, Sr, and Ti. The monochromator lines were used for P (not available on the
318 polychromator), K (very high LOD on the polychromator) and Zr (the monochromator peak
319 was more stable through time). The final analytical precision on the elemental concentrations
320 is therefore lower than 2% (1 sigma) for all elements, except for P (3.60%) and Zr (4.18%).
321 Accuracy for the selected lines was better than 1% for Al, better than 4% for Fe and Ti and
322 could not be calculated for Zr since no reference value exists for PACS-2 (EA-1).
323 In sedimentary geochemistry, elemental ratios are frequently used to overcome dilution by
324 organic and/or biogenic phases (e.g., Van der Weijden., 2002). The theoretical precision on
325 elemental ratios is calculated using Eq. (1).

$$326 \quad RSD_{a/b} = \sqrt{(RSD_a)^2 + (RSD_b)^2}$$

327 where RSD is the relative standard deviation (analytical precision at 1 sigma divided by
328 mean), and a and b are two elements. The precision measured on the elemental ratios obtained
329 with the monochromator is very similar to the theoretical precision. For Ti/Al, for example,
330 the measured precision is 2.25%, whereas the theoretical precision is 2.16%. For the
331 polychromator, however, the measured precision (e.g., 0.68% for Ti/Al) is always much better
332 than the theoretical precision (e.g., 1.22% for Ti/Al), and is also superior to the precision
333 obtained with the monochromator (EA-1). It is also remarkable that, with the polychromator
334 data, the precision of the elemental ratios (Elt/Al) is systematically better than the precision of
335 the corresponding elemental concentrations (EA-1). This increase in precision for elemental
336 ratios is due to the simultaneous acquisition of the elements, which results in the elimination
337 of the signal variability associated with sample preparation and introduction in the plasma,

338 since all spectral lines are affected in similar ways (Schrag, 1999). In consequence, the
339 flickers in signal intensity simply cancel out as elements are detected simultaneously. This
340 demonstrates that ICP-AES is an ideal technique to measure elemental ratios rapidly and at
341 high precision (from 0.1%), provided the instrument is equipped with a
342 polychromator/simultaneous acquisition system.

343

344 **4.2 Calculation of a terrestrial index**

345 One of the main objectives of this study is to define one or several geochemical proxies for
346 terrestrial input/river discharge in the Chilean Fjords. In the modern environment, terrestrial
347 input/river discharge clearly affects (1) surface salinities (Dávila et al. 2002) and (2) the
348 proportion of terrestrial organic carbon in surface sediments (Sepúlveda et al., 2011, Silva et
349 al., 2011). These two variables were therefore used to define a Terrestrial Index (TI), which
350 serves as a reference to evaluate relations between inorganic geochemical results and the
351 energy of the terrestrial river supply. TI is calculated as the score of the first axis of a PCA
352 (Fig. 2) that comprises (1) spring salinity data averaged over the upper 10 m (2, 5 and 10 m
353 depth data, collected during leg 2 of CF7 cruise in November 2001), and (2) the fraction of
354 terrestrial organic carbon (F_{terr}), which is calculated from the N/C ratio of the bulk
355 sedimentary organic matter (Perdue and Koprivnjak, 2007; data in Fig. 3 and Table 2). The
356 inclusion of the F_{terr} data ensures that TI is representative of year-round sedimentation, and it
357 limits the influence of particular salinity values that may be affected by short-term or seasonal
358 changes in precipitation and river discharge (only spring salinity data were available). To
359 calculate F_{terr} , we defined the aquatic end-member as the N/C value of site GeoB3323-4,
360 which was sampled on the continental margin off the coast of Northern Chilean Patagonia
361 (Fig. 1) by Hebbeln et al. (2000) ($N/C = 0.130$, $\delta^{13}C = -19.86$) and the terrestrial end-member
362 as the average of five sediment samples (fraction $<125 \mu m$) collected in rivers that discharge

363 into the fjords ($N/C = 0.0624$, $\delta^{13}C = -27.72$). The N/C value obtained on the river samples is
364 very similar to the average of the seven soil samples ($N/C = 0.0680$) but it is preferred
365 because it is known that N/C ratios are affected by transport in rivers and by weathering in
366 soil profiles (Bertrand et al., 2010). Although bulk sedimentary $\delta^{13}C$ data can also be used to
367 calculate F_{terr} (e.g., Sepúlveda et al., 2011), we prefer using the N/C data because $\delta^{13}C$ values
368 of marine organic matter are affected by several processes such as changes in productivity
369 (e.g., Bickert, 2006). The values of F_{terr} obtained using the two methods are highly positively
370 correlated ($r=0.94$, $p<0.0001$). F_{terr} and salinity (2-10m) are highly correlated ($r=-0.92$,
371 $p<0.0001$, Fig. 3a) and both parameters are independent of sedimentation rates. The resulting
372 TI values, which are linear combinations of F_{terr} and salinity, range between -2.44 (BC1) and
373 1.74 (BC27), with positive values indicating a higher terrestrial input (Fig. 2; Table 2). These
374 values were also used to classify the sedimentary environments in inner- ($TI>0$), mid-
375 ($0<TI<-1$) and outer- ($TI<-1$) fjords (Fig. 2). The main advantage of this index is that it is
376 sensitive to all types of terrestrial inputs, from direct runoff to small streams and large rivers.
377 This is particularly well illustrated by stations 29 and 29A, which are located only 4.3 km
378 from each other but that have clearly distinct TI values (-1.13, and -0.27, respectively)
379 because station 29A is located immediately in front of a small river (Rio Pelu) that discharges
380 into Quitalco fjord.

381

382 **4.3 Sediment composition**

383 *4.3.1 Bulk sediment composition*

384 Surface sediments from the fjords of Northern Chilean Patagonia are composed of lithogenic
385 particles (86.1 ± 6.1 %; calculated as $100\% - \text{bio-opal} - 2.2 \times \text{TOC} - \text{CaCO}_3$, e.g., Nederbragt
386 et al., 2008), biogenic opal (9.0 ± 4.8 %), organic matter (4.2 ± 2.2 %; calculated as $2.2 \times$
387 TOC), and carbonate (0.7 ± 0.9 %) (average ± 1 s.d.; see Table 2 and EA-2 for data).

388 Distribution plots (Fig. 3b) clearly show that biogenic opal is concentrated in the mid- and
389 outer fjords, while carbonate only occurs in the open ocean and outer fjord sites, where it
390 represents carbonate aquatic productivity. TOC concentrations are very variable and reflect
391 the combined contribution of the terrestrial and aquatic sources of organic matter. In
392 consequence, the lithogenic fraction of the sediment is significantly higher in the inner fjords
393 than further towards the open ocean (Fig. 3b). The absence of carbonate in the river and soil
394 sediment samples provides evidence that all CaCO_3 originates from aquatic productivity, in
395 agreement with the geological map (Sernageomin, 2003; Fig. 1). Similarly, the low biogenic
396 opal content of the river and soil sediment samples (1.47 and 2.27 %, respectively, Fig. 3b
397 and EA-2) compared to the fjord sediment samples (9.0 ± 4.8 %) demonstrates that biogenic
398 opal mainly reflects the fjords productivity in siliceous organisms, predominantly diatoms
399 (Rebolledo et al., 2005). The low amounts of biogenic opal measured in the river and soil
400 sediment samples most likely represent a mixture of siliceous organisms thriving in rivers and
401 fresh volcanic glasses that were not entirely accounted for by the detrital Si correction.

402 4.3.2 Mineralogy

403 4.3.2.1. *Bulk mineralogy.* The bulk mineralogical composition of the surface sediment
404 samples (average ± 1 s.d.; data in EA-2; Fig. 4a) is dominated by plagioclase (42 ± 10 %),
405 alkali feldspars (20 ± 7 %), and quartz (15 ± 7 %). Accessory minerals detected by XRD
406 include amphibole (9 ± 8 %), pyroxene (10 ± 6 %), and calcite (4 ± 4 %). Plagioclase was
407 also the dominant mineral in the soil (39 ± 9 %) and river (45 ± 7 %) samples, but alkali
408 feldspar was only detected in 3 soil samples (Burgos 1, Burgos 2, and Trapial 1). Quartz was
409 relatively abundant in the soil (32 ± 25 %) and river (22 ± 15 %) sediment samples, while
410 amphibole was abundant in the river sediment samples (30 ± 12 %) but was below detection
411 limits in the soil samples (Fig. 4a). Calcite was always absent from the river and soil sediment

412 samples and traces of pyroxene were detected in all the soils samples and in some of the river
413 sediment samples (EA-2).

414 The presence in high proportions of plagioclase, alkali feldspar and pyroxene in the soil
415 samples (EA-2) reflects their volcanic origin (Bertrand and Fagel, 2008), in agreement with
416 the soil map of Gut (2008). In addition, the presence of quartz and the absence of amphibole
417 in these samples most likely reflects the alteration of the underlying bedrock which results in
418 the dissolution of amphibole and in the incorporation of quartz grains in the volcanic ash
419 soils. The high amount of feldspars (plagioclase and alkali feldspar) in the two Burgos soil
420 samples ($82 \pm 5 \%$) confirms the dominating volcanic nature of soil parent material in this
421 area, which is frequently affected by explosive eruptions of Hudson, Cay and Maca volcanoes
422 (Chile, 2003; Fig. 1). This interpretation is confirmed by the absence of quartz, and the
423 presence of a broad XRD amorphous diffraction band, which is typical for soils developed on
424 pure volcanic deposits (Bertrand and Fagel, 2008).

425 For the river and fjord sediment samples, the average proportions of plagioclase, alkali
426 feldspar, pyroxene and quartz are representative of a mixture of the volcanic (andosol) and
427 granodioritic sources. The presence of significant amounts of amphibole in these samples
428 reflects the importance of the granodioritic source since amphibole is generally absent from
429 regional volcanic ash deposits (Bertrand and Fagel, 2008) but commonly occurs as
430 Hornblende in the rocks of the North Patagonian Batholith (Nelson et al., 1988; Pankhurst et
431 al., 1999). It was however not possible to quantify the contribution of these two main sources
432 based on our mineralogical results since no quantitative mineralogical data exists for the
433 North Patagonian Batholith.

434 The only minerals that show significant ($p < 0.05$) linear correlations with TI are alkali
435 feldspar ($r = -0.57$), which increases towards the open ocean, and calcite ($r = -0.65$), which
436 only occurs in the open ocean and outer fjord samples, in agreement with the carbonate

437 content data. Amphibole and quartz also decrease towards the open ocean (Fig. 4a), but these
438 relations are not significant at $p < 0.05$. An explanation for the concentration of quartz and
439 amphibole in the inner fjords samples is their refractory and density properties (Galy and
440 France-Lanord, 2001, Nebsitt and Young, 1996), i.e., quartz is a relatively refractory mineral
441 that is frequently concentrated in the coarse fraction of the sediment and amphibole is
442 relatively dense and can therefore not be transported on long distances. Therefore, only some
443 small grains of these two minerals can be transported to the Pacific Ocean by the CFW.

444

445 *4.3.2.2. Clay mineralogy.* The clay mineralogy of the fjord samples is dominated by illite (48
446 ± 14 %) and chlorite (27 ± 9 %), with minor amounts of smectite (12 ± 14 %) and kaolinite
447 (13 ± 7 %) (average ± 1 s.d.; data in EA-2). These proportions seem to be typical for clay
448 minerals at the mid- to high latitudes of the eastern south Pacific (Fütterer, 2006). The
449 dominance of illite and chlorite is explained by physical erosion of the biotite-rich North
450 Patagonian batholith. The high variability in the smectite content of the fjord sediments (0 to
451 36 %) most likely originates from regional variations in volcanism, since smectite typically
452 originates from the low temperature alteration of volcanic products (Fütterer, 2006). Illite
453 decreases ($r = 0.55$, $p = 0.04$) and smectite increases ($r = -0.62$, $p = 0.02$) towards the Pacific
454 Ocean. Kaolinite and chlorite do not show any significant trend.

455

456 *4.3.3 Grain size*

457 The grain size of the lithogenic fraction shows a general decreasing trend towards the open
458 ocean (Fig. 4b; data in Table 2 and EA-2). The only two exceptions are the fine-grained
459 sediments at site 27, which is located in front of a pro-glacial river and therefore mainly
460 receives fine glacial clays and silts, and the sand-dominated sample from site 1, which is
461 located at a depth of 240 m in front of Guafo Island, where strong currents are frequent

462 (Sievers and Silva, 2008), and where steep slopes may result in the development of mass-
463 wasting deposits. The relation between grain size and TI is very well expressed in the
464 restricted dataset ($r=0.76$, $p=0.08$).

465

466 *4.3.4 Inorganic geochemistry*

467 *4.3.4.1 Elemental concentrations and selection of an element representative of the lithogenic*

468 *fraction.* Among the thirteen measured geochemical elements, only Al is positively correlated
469 with the lithogenic fraction of the sediment at $p<0.05$ ($r=0.90$, $p<0.001$; EA-4). This
470 observation implies that the Al content of the lithogenic fraction of the sediment is almost
471 constant for the entire fjord region (Fig. 4a). It does not vary with TI ($r=0.03$; Fig. 4) or with
472 grain size ($r=0.04$). In addition, the Al concentration of the river sediment samples ($8.09 \pm$
473 0.48 %) varies little and is similar to the Al concentration of the North Patagonian Batholith
474 (8.16 ± 1.16 %, data from Pankhurst et al., 1999) and of the Quaternary volcanic rocks ($8.87 \pm$
475 0.63 %, data from Naranjo and Stern, 1998 and D'Orazio et al 2003). The only source that has
476 higher Al concentrations is the regional volcanic soils (9.25 ± 0.98 %), where Al is
477 concentrated during pedogenesis (preferential dissolution of K, Na, Mg and Ca-rich minerals;
478 Nesbitt and Young, 1989). The similar Al concentrations of the fjord, river and regional rock
479 samples is due to the lithogenic and immobile nature of Al, and its presence in similar
480 concentrations in most igneous and metamorphic rock-forming minerals (Calvert et al., 2001;
481 McLennan et al., 2003). The difference with the soil samples most likely indicates that the
482 contribution of soil material to sedimentation in the fjords is relatively minor. Al is therefore
483 relatively insensitive to changes in the nature of sediment sources, catchment size, and
484 hydrodynamic sorting. As a result, Al is the ideal element to represent the lithogenic fraction
485 of the sediment, and to use as a normalizer for other lithophile elements. Although Fe, Ti and
486 Zr are also generally considered as lithophile and immobile elements (McLennan et al., 2003),

487 they only show weak and statistically insignificant positive linear correlations with the
488 lithogenic content of the sediment ($r=0.22$ to 0.39).

489

490 *4.3.4.2 Geochemical ratios, Terrestrial Index and grain size.* In the following we analyze
491 changes in the composition of the lithogenic fraction of the sediment by examining the
492 behavior of Al-based elemental ratios (El/Al) and Ti/Fe, Zr/Fe, Zr/Ti. We specifically
493 evaluate the relations between these elemental ratios and TI and grain size (Fig. 5).

494 None of the elemental ratios are linearly correlated with TI at $p<0.05$ (EA-4). The only two
495 weakly significant correlations are with Ca/Al ($r=-0.51$, $p=0.06$) and Si/Al ($r=-0.51$, $p=0.06$).
496 However, these two negative correlations do not reflect a change in the lithogenic fraction,
497 but they are due to the global increase in carbonate (Ca/Al) and biogenic silica (Si/Al)
498 productivity towards the outer fjords. Although the composition of the lithogenic fraction is
499 not linearly correlated with TI, elemental ratios Fe/Al, Ti/Al and Zr/Al seem to increase and
500 subsequently decrease towards the marine environment (Fig. 5), with the slope of the
501 decreasing side of the paraboloid being steeper for Zr/Al than for Ti/Al, which is itself steeper
502 than for Fe/Al.

503 Grain size, on the other hand, is strongly positively correlated with litho-Si/Al ($r=0.74$,
504 $p=0.003$; EA-4), which supports the interpretation that quartz is concentrated in the coarse
505 fraction of the sediment (see 4.3.2.1). Grain size is also negatively correlated with Fe/Al
506 ($r=-0.71$, $p=0.005$), which implies that Fe-rich minerals are concentrated in the fine fraction
507 of the sediment. The other elemental ratios do not show significant linear correlations with
508 grain size.

509

510 *4.3.5. Magnetic susceptibility*

511 In the surface sediment samples, magnetic susceptibility (MS) displays strong variations that
512 are remarkably parallel to the grain size of the lithogenic fraction of the sediment ($r=0.93$,
513 $p<0.0001$; Figs 4b and 6; data in EA-4). A few geochemical elements are significantly
514 correlated with MS, but the most significant correlation is with litho-Si/Al ($r=0.67$, $p=0.009$),
515 which confirms that the MS of Chilean fjord sediments is mainly driven by grain size. This
516 high correlation is confirmed by the restricted dataset ($r=0.86$, $p=0.03$ for MS vs litho-Si; EA-
517 5). This restricted data set also displays a highly significant correlation between MS and the
518 lithogenic fraction of the sediment ($r=0.89$, $p=0.02$), between MS and grain size ($r=0.91$,
519 $p=0.01$). As a consequence, MS is positively correlated to the combination of grain size and
520 the lithogenic fraction of the sediment ($r=0.97$, $p=0.01$; Fig 6b). In the restricted dataset, MS
521 is also highly correlated with TI ($r=0.86$, $p=0.03$).

522 We assume that the relation between MS and the combination of grain size and the lithogenic
523 fraction of the sediment is due to (1) signal dilution by non-lithogenic particles, and (2)
524 enrichment of heavy ferromagnetic minerals in the coarse fraction of the sediment during
525 sedimentary sorting (McLennan et al. 2003). An alternative hypothesis for the significant
526 correlation between MS and grain size is the presence of micro-inclusions of magnetite
527 (Scofield and Roggenthen, 1986) in refractory, and therefore coarse, minerals usually
528 considered as paramagnetic. Although Andrews (2008) observed a correlation between some
529 paramagnetic minerals and MS in the Denmark Strait, more specific analyses need to be
530 performed to refine this relation.

531

532 *4.3.6 Principal component analysis*

533 The results of the PCA on the full dataset show a complex structure, with the first two PCA
534 axes accounting for a mere 51.07 % of the total variance (Fig. 7a). The first axis primarily
535 reflects variance in (1) Si and Litho-Si, (2) Al and the lithogenic fraction, and, to a lesser

536 extent, (3) Zr, Ti/Fe, Zr/Fe, and Zr/Ti. The loadings of TI and grain size on PCA axis 1 are
537 low, but these two variables load nearly equally on both F1 and F2. Calcite, $\delta^{15}\text{N}$, $\delta^{13}\text{C}$,
538 Ca/Al, and the carbonate content of the sediment load negatively on both F1 and F2, in
539 opposite direction of TI. These variables are associated with salinity and depth, which are
540 indicative of an increased marine influence. Fig 7a also depicts the combined influence of TI
541 and grain size on a series of variables, such as Ti/Fe, Zr/Fe, Zr/Ti, and pyroxene. The PCA
542 biplot also displays the relatively tight grouping of Litho-Si, Litho-Si/Al, quartz and grain
543 size, which confirms that grain size exerts a strong control on these three geochemical
544 variables. Similarly, a tight grouping is observed for Ti, Zr/Al, and TI. Except for P and P/Al,
545 which are indicators for the presence of the relatively dense mineral apatite, no variable has a
546 strong F2 loading.

547 On the restricted dataset, the first two PCA axes account for a cumulative 73.22 % of the
548 variance, with PCA axis 1 explaining nearly half (47.77%) of the total variance. The PCA
549 biplot of the restricted dataset (Fig. 7b) shows a similar pattern to the PCA biplot of the full
550 dataset, but with much higher F1 loadings for TI and grain size (0.78 and 0.79, respectively).
551 This results in a tighter grouping of the variables controlled by these two parameters, such as
552 Ti/Fe, but also Litho-Si, Zr, Si, Litho-Si/Al and Zr/Al, with respective F1 loadings of 0.99,
553 0.98, 0.96, 0.95, and 0.91. The PCA biplot of the restricted dataset (Fig. 7b) also shows that
554 PCA axis 2 captures most of the variance in Ti, Fe, Ti/Al, Mg, and amphibole, which are
555 typical indicators for mafic minerals, and P and P/Al, which represent apatite. These results
556 demonstrate that the grain size of the sediment and its content in mafic, i.e. dense, minerals
557 are independent variables, although both seem to co-vary with TI.

558

559

5. DISCUSSION

560 The main factors that control the mineralogy and geochemical composition of siliclastic
561 sediments are (1) the nature of the provenance, (2) the intensity of physical and chemical
562 weathering and (3) the processes that occur during sediment transport (Nesbitt and Young,
563 1996). Here, we assess the importance of these three factors for the fjords of Northern Chilean
564 Patagonia.

565

566 **5.1 Provenance**

567 Our mineralogical data suggest that the fjord sediments originate from varying mixtures of (1)
568 volcanic particles from the andosols and (2) minerals from the regional bedrock, which is
569 dominated by granitoids and tonalities of the North Patagonian Batholith (Pankhurst et al.,
570 1999; Parada et al., 2007). In addition, secondary lithologies, such as the Quaternary and
571 Mesozoic volcanic rocks and the volcano-sedimentary rocks of the Traiguen Formation (Fig
572 1) likely play a role in the supply of sediment to some of the fjords. Since these lithologies are
573 not equally represented in all the watersheds, it is reasonable to expect spatial variations in
574 provenance. However, the results obtained on our river sediment samples demonstrate that the
575 particles that are discharged to the fjords have a relatively constant geochemical composition
576 (e.g., Al: 8.09 ± 0.48 %; EA-2) that is in agreement with a mixture of the regional volcanic
577 (Al: 8.87 ± 0.63 %) and granitoidic (Al: 8.16 ± 1.16 %) sources. The low variability in the
578 concentration of other elements in the river sediment samples (e.g., Fe: 6.57 ± 0.82 %; Ti:
579 0.83 ± 0.09 %) confirms the relative homogeneity of the sediment particles supplied to the
580 fjords. This relative homogeneity is explained by the effective mixing of the source rocks and
581 soils during river transport, which results in the smoothing of the relatively variable chemical
582 compositions (e.g., Gaillardet et al., 1999). As a result, the nature of the sediment supplied to
583 the fjords of Northern Chilean Patagonia is relatively independent of regional variations in
584 lithology, drainage areas, soil thickness, volcanic influence, etc.

585 It is worth noting that drainage areas, soil thickness, volcanic activity, and glacier cover most
586 likely varied during the Holocene. Although modern spatial variations of these parameters do
587 not seem to significantly affect provenance, the amplitude of change of these parameters
588 during the Holocene is relatively unknown. The possibility that provenance changed during
589 the Holocene can therefore not be discarded, and it should be taken into account when
590 interpreting data from long fjord sediment cores. Additional long-distance sources of
591 sediment transported by aeolian processes may also have significantly affected provenance in
592 drier and windier climate conditions.

593

594 **5.2 Weathering**

595 The presence in important proportions of minerals highly susceptible to chemical weathering
596 such as plagioclase, amphibole and pyroxene in the river and fjord sediment samples (EA-2)
597 demonstrates that the sources of sediment to the fjords are mostly fresh. Low chemical
598 weathering is also confirmed by (1) the clay mineralogy, which is dominated by minerals
599 characteristic of physical weathering (illite and chlorite); and by (2) the low values of the
600 Chemical Index of Alteration (CIA; Nesbitt and Young, 1982) and Chemical Index of
601 Weathering (CIW; Harnois, 1988) calculated for the surface (CIA: 48.1; CIW: 44.0; Table 2)
602 and river sediment samples (CIA: 46.4; CIW: 48.2; Table 2, Fig. 8). These values are typical
603 for fresh material (≤ 50 ; fully weathered materials would have values of 100; Price and
604 Velbel, 2003) and are nearly identical to the values calculated for the granitoids of the North
605 Patagonian Batholith (CIA: 48.8; CIW: 54.3; data from Pankhurst et al., 1999). Low chemical
606 weathering in Northern Patagonia is due to the combined effect of (1) the characteristic cold
607 climate of the region (Nesbitt and Young, 1996); (2) the regional lithology, i.e. granitoids is
608 one of the less reactive lithologies (White and Blum, 1995); and (3) the recent deglaciation of
609 the region, which results in limited time for chemical weathering of the bedrock.

610 By comparison, the CIA and CIW indices of the soil samples are significantly higher (CIA:
611 66.7; CIW: 71.0; Table 2, Fig. 8), which demonstrates that chemical weathering depleted the
612 soils in the most soluble elements (Ca, Mg, Na and K), and enriched them in residual
613 elements such as Al. The low values of the CIA and CIW indices for the river and fjord
614 surface sediment samples compared to the soil samples (Fig. 8) is therefore an additional
615 argument that demonstrates that the soils contribute little to provenance, in agreement with
616 the mineralogical and Al concentration data. This statement is also supported by the A-CN-K
617 and A-CN-K-FM triangular plots (Nesbitt and Young, 1989) presented in Fig. 8.

618 The intensity of chemical weathering most likely varied during the Holocene. However the
619 continuously cold climate of Northern Chilean Patagonia since the last deglaciation (e.g.,
620 Kaiser et al., 2005) most likely favored physical over chemical weathering, as higher
621 temperature is needed to increase the rate of chemical weathering. More intense physical
622 weathering, and therefore a higher supply of unaltered minerals, during the deglaciation is
623 very likely. The sediment particles that reach the fjords are therefore mainly fresh, and the
624 distribution of soluble elements such as Na, Mg and K in the fjord sediments mainly reflects
625 mineralogical sorting.

626

627 **5.3 Mineralogical sorting during sedimentary transport**

628 Although inorganic geochemical records are frequently interpreted in terms of past changes in
629 terrestrial supply (e.g., Haug et al., 2001; Lamy et al., 2004), there is still an evident lack of
630 understanding of the effect of physical processes, mainly sediment transport, on sediment
631 mineralogy and geochemistry. One of the most important studies in this respect is by Nesbitt
632 and Young (1996), who studied the composition of sediments deposited in the fluvio-glacial
633 system of Guys Bight, Baffin Island. These authors found that the primary mafic minerals
634 (olivine, pyroxene, amphibole and biotite) of the bedrock, and therefore Mg and Fe, are

635 enriched in fine sands and muds (i.e., silt + clay), while the coarse sands contain a greater
636 amount of Si. According to Rosenbaum and Reynolds (2004), this distribution probably
637 reflects the small grain size of the heavy mafic minerals of Baffin Island, which makes them
638 hydrodynamically equivalent to less dense minerals such as quartz and feldspar.

639 Our results from the fjords of Northern Patagonia seem to confirm that grain size and mineral
640 density vary independently, but they also show that both variables are partly related to TI.

641 In addition, our results show that, among the four lithogenic elements of interest (Al, Fe, Ti,
642 Zr), Al is the only element that clearly reflects the lithogenic content of the sediment. Its
643 distribution is not influenced by grain size or grain density, which means that Al, as a total
644 sum, remains constant in the lithogenic fraction of the fjord sediments. It is therefore
645 independent of sedimentary transport processes. The distribution of Zr, Ti and Fe in the
646 Chilean fjords, however, seems to be controlled by their association with heavy and/or coarse
647 minerals.

648 Iron is generally associated with relatively dense mafic minerals (amphibole, pyroxene,
649 olivine; $d=2.9-3.5$), although it occurs in minor proportions in a large series of minerals,
650 including clays (e.g., Monroe and Wicander, 2009). It is absent from minerals that result from
651 intense weathering, such as kaolinite, which is rare in the Chilean fjords. In environments rich
652 in peat, Fe can also be leached from the peat and transported to sedimentary environments
653 fixed on Dissolved Organic Carbon (DOC), where it can precipitate as iron-hydroxides
654 (Krachler et al., 2010). In Northern Chilean Patagonia, this source is negligible since peat
655 bogs are much more common south of 48°S (Gut, 2008), and DOC values of North
656 Patagonian streams (0-6 mg/l; Perakis and Hedin, 2002) are one order of magnitude lower
657 than DOC values measured in rivers flowing out of peat bogs (40-70 mg/l; Lal et al., 1997).

658 Our data show that the distribution of Fe with TI is rather complex (Fig. 5a). The Fe/Al ratio
659 is highest in the inner-fjords with $0 < \text{TI} < 1$, and it decreases towards the main land ($\text{TI} > 1$), as

660 well as towards to open ocean ($TI < 0$). The distribution of amphibole with TI (Fig. 4) being
661 similar to the distribution of Fe/Al with TI, and the close association between amphibole,
662 Mg/Al and Fe/Al in the PCA biplots (Fig. 7) confirm that the distribution of Fe is driven by
663 the concentration of mafic minerals in the mid-fjords. Our results also show that Fe is
664 concentrated in the silt fraction of the sediment (Fig. 5b), in agreement with the observations
665 of Nesbitt and Young (1996) in Guys Bight, and probably due to the low refractoriness of
666 mafic minerals. These data therefore demonstrate that Fe and Mg in the Chilean fjords are
667 associated with sediment particles of low to intermediate grain size and intermediate density.
668 The low concentrations of Fe and Mg in some of the most proximal sites most likely reflect
669 dilution by coarser and/or denser minerals such as quartz. The occurrence of Fe in open
670 marine samples is likely due to its presence in very fine-grained mafic minerals, in
671 plagioclase (density: 2.6-2.7), and in most clay minerals.

672 Zirconium is most often associated with zircon, which is a typical accessory mineral of the
673 North Patagonian Batholith. Zircon is a very dense ($d = 4.6-4.7$) and refractory mineral, which
674 results in Zr being concentrated in the densest and coarsest mineralogical fraction of
675 sediments. In Chilean Patagonia, Zirconium concentrations are particularly high in the coarse
676 soil and river sediment samples. In the fjord sediments, Zr and Zr/Al show a strong
677 association with TI (Fig. 7), which results from its relation with both grain size and density
678 (Fig. 5c). The Zr/Al ratio of sediments is therefore maximum where both grain size and grain
679 density are high, i.e. in the most proximal environments, where the energy of river discharge
680 is maximum. It quickly decreases towards the open ocean (Fig. 5c).

681 Titanium is frequent in igneous rock-forming minerals (Verhoogen, 1962). It occurs in most
682 mafic minerals (amphibole, pyroxene, olivine; $d = 2.9-3.5$; Nesbitt, 2003) and in ilmenite, in
683 association with iron, and it is a main constituent of less frequent and relatively dense iron-
684 free minerals such as rutile and titanite (Verhoogen, 1962, Nesbitt and Young, 1996,

685 McLennan et al. 2003; $d=3.5-4.8$). Ti is therefore associated with minerals of refractoriness
686 (i.e., grain size) and density similar or higher than Fe-bearing minerals, but lower than Zr-
687 bearing minerals. A direct consequence for sedimentary environments is that the distribution
688 of Ti is similar to that of Fe, although it is transported on smaller distances than the average
689 Fe-bearing minerals. Ti is associated with minerals that are transported on longer distances
690 than zircon (Fig 5d). This is also observed in our dataset, where Ti and Ti/Al are in
691 intermediate position between Fe and Zr (Fig. 5). The behavior of Ti in very proximal
692 (deltaic) environments is not well resolved in our samples but because of its association with
693 mafic minerals, Ti concentrations are expected to decrease in the very coarse and very dense
694 fraction of the sediment. This relation should however be confirmed by studying samples
695 collected in proximal environments at high spatial resolution.

696 The consequences of these element-mineral associations for the distribution of Al, Fe, Ti and
697 Zr in Chilean fjord sediments are schematically represented in figure 9. Proximal locations,
698 such as deltaic environments, contain high amounts of Zr, low amounts of Fe, and
699 intermediate amounts of Ti. The concentrations in Zr quickly decrease towards more distal
700 locations, while the Fe concentrations increase. The inner fjords are characterized by
701 increasing amounts of Ti and Fe. Ti concentrations peak in the mid-fjords and then rapidly
702 decrease towards the open ocean. The highest concentrations in Fe occur at the limit between
703 the mid and outer fjords. In the outer fjords and on the continental margin, the concentrations
704 in Zr and Ti are minimal, and the Fe concentration decreases with distance from the
705 tributaries.

706

707 **5.4 Proxies of terrestrial sediment discharge**

708 The distribution of the inorganic geochemical elements Al, Fe, Ti, and Zr in the Chilean
709 fjords (Fig. 9) has important implications for inorganic geochemical proxy-records of
710 terrestrial sediment discharge:

711 (1) Measuring Al-based elemental ratios at high precision is of utmost importance to
712 accurately calculate variations in the inorganic geochemical composition of terrestrial
713 sediments. For discrete samples, simultaneous acquisition ICP-AES technology is ideal.
714 Although this technique allows the measurement of long sediment cores at reasonable
715 resolution, XRF core scanning is becoming the technique of choice to generate geochemical
716 records at very high resolution. XRF core scanners, however, are limited by their low
717 accuracy for light elements, such as Al (Tjallingii et al., 2007). Technical efforts should
718 therefore be put forward to increase the analytical precision of XRF core scanners for light
719 elements. Substituting Al by Ti or other lithophile elements such as Rb (e.g., Rothwell et al.,
720 2006) may lead to a biased interpretation.

721 (2) The elemental ratios Fe/Al, Ti/Al and Zr/Al are well suited for estimating changes in the
722 energy of river discharge into the fjords through time (Fig. 9).

723 (3) Not all elemental ratios are sensitive in all environments (Fig. 9). For example, deltaic
724 environments are more sensitive to changes in Zr/Al, than more distal environments, such as
725 outer fjords and continental margins, where Fe/Al is especially useful.

726 (4) Caution should be exercised when interpreting Fe/Al data in terms of past river discharge,
727 particularly in proximal environments. The association of Fe with minerals of intermediate
728 grain size and intermediate density results in the non-linearity between Fe/Al and the intensity
729 of river discharge. This relation, which is frequently assumed to be linear when interpreting
730 sediment core records, is only valid for distal locations, where sediments are relatively fine-
731 grained.

732 The use of elements other than Al, Fe, Ti and Zr to reconstruct past changes in terrestrial
733 supply is less straightforward. Our data demonstrate, for example, that Mg/Al is closely
734 associated to mafic minerals (r with Fe/Al = 0.91, $p < 0.001$), which indicates that Mg/Al could
735 also be used to reconstruct changes in hydrodynamic conditions in distal environments.
736 However, since Mg is also associated to ocean carbonate productivity (e.g., Raitzsch et al.,
737 2010), this relationship may not be valid where carbonate productivity is high. In addition,
738 our data demonstrate that the litho-Si and quartz contents of the sediment can be used as
739 proxies for grain size. Their use is however restricted to low resolution records since they
740 require either the analysis of discrete samples by XRD, or the measurement of both total Si
741 and bio-Si, which is time-consuming.

742 In addition to controlling the inorganic geochemical composition of sediments, mineralogical
743 sorting processes also affect other sediment properties, such as magnetic susceptibility. For
744 the Chilean Fjords, our data suggest that MS can be used as a first-order indicator of
745 hydrodynamic changes, since MS is primarily controlled by the grain size of the lithogenic
746 content of the sediment. At sites where the concentration of organic and biogenic particles is
747 high and potentially variable, MS data should be normalized to the lithogenic fraction (or Al)
748 before being used as a proxy for grain size.

749 Finally, our data confirm that the bulk organic geochemical composition of the sediment
750 (C/N, $\delta^{13}\text{C}$, $\delta^{15}\text{N}$) remains a powerful tool for assessing terrestrial sediment supply. One of the
751 main advantages of this approach is that it allows the estimation of the relative proportions of
752 marine and terrestrial organic carbon preserved in sediments (e.g., Perdue and Koprivnjak,
753 2007). Organic-based proxies are however not applicable to sediments with very low TOC,
754 such as those deposited in glacio-marine environments.

755

756 **5.5. Application to sediment cores**

757 To test the validity of our geochemical proxies for paleo-reconstructions, we investigated
758 variations in Fe/Al, Ti/Al and Zr/Al in two long sediment cores collected in two different
759 environments in the fjords of Northern Patagonia. Core JPC14 was collected in front of a pro-
760 glacial river (Fig. 1) and represents an inner-fjord site (TI = 1.74), while core PC29A was
761 collected in the center of Quitalco fjord (TI = 0.-27), and is therefore characteristic of a mid-
762 fjord site (Fig. 9). The results (Fig. 10) demonstrate that at site JPC14, Zr/Al is the most
763 sensitive elemental ratio (higher variability, expressed as RSD), while for core PC29A, Ti/Al
764 is the most sensitive. Figure 10 also clearly demonstrates that at the inner fjord site (JPC14),
765 Zr/Al increases and Fe/Al decreases with increasing hydrodynamic energy, while Ti/Al
766 remains rather insensitive. In core PC29A (mid-fjord site), Zr/Al is rather insensitive, while
767 both Ti/Al and Fe/Al decrease with increasing hydrodynamic energy. These data therefore
768 support our findings that Zr/Al and Ti/Al are the most sensitive elemental ratios in the inner
769 fjords and mid-fjords, respectively. They also confirm that in these proximal environments,
770 Fe/Al is always inversely related to grain size and therefore to hydrodynamic energy.
771 Furthermore, these results show that the relation between Zr/Al, Ti/Al, Fe/Al and
772 hydrodynamic energy is valid for at least the last 5400 yrs. Also, the concentrations in the
773 soluble elements Ca, Na and K in the sediment core samples do not change significantly
774 through time, as they suggest that chemical weathering in Northern Patagonia was very
775 limited during the last 5400 years (Fig. 8). Similar low chemical weathering conditions are
776 expected during the deglaciation (17-12 kyr BP; Hulton et al., 2002) since the climate was
777 colder and glaciers were more expanded than during the Neoglaciation (~4500-1000 BP;
778 Glasser et al., 2004). Although temperature was ~2°C higher during the early Holocene
779 (Heusser and Streeter, 1980; Glasser et al., 2004), this was likely not sufficient to significantly
780 increase chemical weathering rates (Gislason et al., 2009), especially since precipitation
781 decreased concomitantly (Heusser and Streeter, 1980). Given that chemical weathering likely

782 remained low over the deglaciation and Holocene, the proposed proxies of energy of the
783 terrestrial sediment discharge should therefore be valid for the entire period of time recorded
784 by Northern Patagonia fjord sediments.

785

786

6. CONCLUSIONS

787 The inorganic geochemical composition of the sediments deposited in the fjords of Northern
788 Chilean Patagonia is primarily controlled by hydrodynamic mineralogical sorting. Among the
789 four typical lithogenic and mainly immobile elements that are commonly used for the study of
790 past changes in terrestrial input (Al, Fe, Ti, Zr), only Al is independent of hydrodynamic
791 processes. Its concentration reflects the proportion of lithogenic particles, which supports the
792 common practice of normalizing other lithophile elements by Al for assessing changes in the
793 composition of the lithogenic fraction. The distribution of Fe, Ti and Zr is controlled by their
794 association with heavy and/or coarse minerals. Our results show that the sensitivity of Zr/Al,
795 Ti/Al and Fe/Al to changes in the energy of the terrestrial supply to the fjords varies with
796 distance from the tributaries. Zr/Al and Ti/Al are most useful in deltaic and proximal fjord
797 environments, respectively, while increases in Fe/Al can only be translated into higher
798 hydrodynamic conditions in distal environments, such as outer fjords and continental margins.
799 Caution should be exercised when using Fe/Al in proximal environments, where it shows an
800 inverse relation with the energy of the terrestrial supply. In addition, our data suggest that MS
801 can be used as a first-order indicator of changes in grain size and in the relative proportion of
802 lithogenic particles.

803 Finally, this dataset constitutes a strong basis for the interpretation of future sedimentary
804 records from the fjords of Chilean Patagonia in terms of past climate and environmental
805 change. The application of the inorganic geochemical proxies developed in this study to long
806 sediment cores should provide important information regarding past changes in river sediment

807 discharge, which is primarily controlled by precipitation and glacier melting. The principles
808 derived from this study may also be applicable to other high-latitude sedimentary basins
809 dominated by inorganic terrestrial inputs.

810

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1082 **Table captions**

1083

1084 Table 1 – Location of sampling stations

1085 ^a Average of measurements at 2, 5 and 10 m water depth obtained during the second leg of

1086 CF7 cruise in November 2001

1087 ^b from Hebbeln et al. (2000)

1088

1089 Table 2 – Geochemical, mineralogical and sedimentological data used in figures 2 to 6.

1090 ^(a) TI = Terrestrial Index

1091 ^(b) Salinity data from leg 2 (spring) of cruise CF7 (November 2001) - data are available on the

1092 CENDHOC website (http://www.shoa.cl/n_cendhoc/)

1093 ^(c) F_{terr} = Fraction of terrestrial organic carbon. See section 4.2 for details.

1094 ^(d) GS = Mean grain size

1095 ^(e) MS = Magnetic Susceptibility

1096 ^(f) Fe in litho. = Percentage of iron in the lithogenic fraction

1097 ^(g) Al in litho. = Percentage of aluminum in the lithogenic fraction

1098 ^(h) CIA = Chemical Index of Alteration (Nesbitt and Young, 1982), calculated as

1099 $[Al_2O_3/(Al_2O_3+CaO+Na_2O+K_2O)]*100$, using molecular proportions and after removal of

1100 carbonate CaO and salt Na₂O

1101 ⁽ⁱ⁾ CIW = Chemical Index of Weathering (Harnois, 1989) calculated as

1102 $[Al_2O_3/(Al_2O_3+CaO+Na_2O)]*100$, using molecular proportions and after removal of

1103 carbonate CaO and salt Na₂O

1104 ⁽ⁱ⁾ s.d. = standard deviation (1 sigma)

1105

1106 **Figure captions**

1107

1108 Figure 1 – Location of the sampling sites on a simplified geological map of Northern Chilean
1109 Patagonia. The lithological units were drawn according to Sernageomin (2003). Superficial
1110 Quaternary deposits were omitted. The bathymetric data are from NDGC-NOAA. The border
1111 between Chile and Argentina follows the drainage divide. Sediment core PC29A was
1112 collected at the same site than surface sample BC29A. NPI = Northern Patagonian Ice Field

1113

1114 Figure 2 – Results of the principal component analysis (PCA) used to calculate the Terrestrial
1115 Index (TI = score of PCA axis 1) from the Fraction of Terrestrial Carbon (F_{terr}) and salinity
1116 data. TI can be calculated for any site using the equation indicated on top of the plot, with
1117 F_{terr} in % and salinity in psu. This equation provides TI values of 6.4 and -4.2 for fully
1118 terrestrial (salinity: 0‰; F_{terr} 100%) and fully marine (salinity: 35‰; F_{terr} 0%)
1119 environments, respectively.

1120

1121 Figure 3 – Spatial variability of indicators of terrestrial input (a), and bulk sediment
1122 composition (b). The surface sediment samples are organized according to their terrestrial
1123 index (TI) in order to visualize the compositional variations along a proximal to distal
1124 transect: inner fjord ($TI > 0$), mid-fjord ($0 < TI < -1$) and outer fjord ($TI < -1$). Data are
1125 presented in Table 2. See sections 4.2 and 4.3 for details.

1126 a) Terrestrial index (TI), salinity (0-10 m), and fraction of terrestrial carbon (F_{terr} , %)

1127 b) Bulk sediment composition, expressed as percentages of biogenic opal, total organic
1128 carbon (TOC), carbonate and lithogenic fraction.

1129 ¹ The open ocean data correspond to site GeoB3323-4 off the coast of Southern Chile

1130 (Hebbeln et al., 2000; Romero et al., 2001).

1131

1132 Figure 4 – Spatial variability of selected variables. The samples are organized as in Figure 3
1133 and the data are presented in Table 2.

1134 a) Selected minerals (amphibole, quartz) and geochemical elements (Fe, Al)

1135 b) Grain size and magnetic susceptibility. The magnetic susceptibility values are primarily
1136 controlled by the grain size and the lithogenic content of the sediment. See section 4.3.5

1137 ¹ The open ocean data correspond to site GeoB3323-4 off the coast of Southern Chile (Lamy
1138 et al., 1998; Klump et al., 2000). The grain size data of Lamy et al. (1998) were not used since
1139 these authors analyzed the silt fraction (2-63 μm) only.

1140

1141 Figure 5 – Relationships between elemental ratios (Fe/Al, Ti/Al and Zr/Al, in g/g) and grain
1142 size and the Terrestrial Index for the full dataset. The trends shown in red in (a) do not
1143 correspond to any mathematical function and have therefore no statistical value. Note that the
1144 grain size scale on the Zr/Al plot is inverted for clarity. The correlation coefficients (r) apply
1145 to paraboloids fitting through the points. The slope towards negative TI values (distal
1146 environment) is steeper for Zr/Al than for Ti/Al, which is in turn steeper than for Fe/Al.

1147

1148 Figure 6 – 3-D correlation plot of magnetic susceptibility vs. lithogenic content of the
1149 sediment and grain size. (a) Full dataset; (b) Restricted dataset (Jacaf and Puyhuapi fjords).

1150

1151 Figure 7 – Principal component analysis (PCA) biplot showing the relationships between the
1152 52 measured variables. (a) Full dataset (n=14); (b) Dataset restricted to samples from Jacaf
1153 and Puyhuapi fjords (n=6). The green and red colors highlight the variables that are controlled
1154 by mineral density (Fe, Fe/Al, Ti, Ti/Al, Mg, Mg/Al, P, P/Al, amphibole, and pyroxene), and
1155 grain size (mean grain size, quartz, Si, Si/Al, Litho-Si, Litho-Si/Al, Zr, Zr/Al, Zr/Fe, Zr/Ti,

1156 Ti/Fe and magnetic susceptibility), respectively. Brown represents more terrestrial and blue
1157 more marine sediments. Black dots correspond to variables with no clear affinity. The
1158 variables appear better grouped on the plot corresponding to the restricted dataset (b), where
1159 specific data clusters can clearly be identified. The plots demonstrate that the variables related
1160 to grain size and mineral density are not related but that they are both partly controlled by TI.
1161 See section 4.3.6 for details.

1162 ¹: Carbonate content measured by coulometry; ²: Calcite wt. % measured by x-ray diffraction.
1163

1164 Figure 8 – Ternary plots A-CN-K and A-CNK-FM where A=Al₂O₃, C=CaO (silicate fraction
1165 only), N=Na₂O, K=K₂O, F=total Fe as FeO, and M=MgO (in mole fraction, following Nesbitt
1166 and Young, 1989). The weathering trends and the granodiorite and andesite areas are from
1167 McLennan et al. (1993). The NPB (North Patagonian Batholith) data are from Pankhurst et al.
1168 (1999). CIA = Chemical Index of Alteration (Nesbitt and Young, 1982). The two plots
1169 demonstrate that the composition of the river and fjord sediment samples is relatively
1170 homogenous, and is similar to the composition of the rocks forming the North Patagonian
1171 Batholith. They also demonstrate that the contribution of the regional andosols is minor. Data
1172 are presented in Table 2 and EA-2.

1173
1174 Figure 9 – Schematic representation of the distribution of Al, Fe, Ti and Zr in the lithogenic
1175 fraction of the sediment, in relation to distance to the tributaries and energy of the river
1176 supply. This illustration demonstrates that not all elemental ratios are sensitive in all types of
1177 fjord sedimentary environments. The location of cores JPC14 and PC29A (see Fig. 10) is also
1178 indicated. The scale of the vertical and horizontal axes is arbitrary.

1179

1180 Figure 10 – Normalized elemental ratios (g/g) versus grain size for sediment cores JPC14
1181 (Golfo Elefantas) and PC29A (Quitralco fjord). Grain size is used here as an indicator of
1182 hydrodynamic conditions. The relative standard deviation of the normalized data is used to
1183 describe the variations between extreme conditions. The determination coefficients (r^2) apply
1184 to power curves fitting through the points. This figure confirms that Zr/Al and Ti/Al are the
1185 most sensitive elemental ratio in the inner fjords and mid-fjords, respectively. Data are
1186 presented in EA-3.

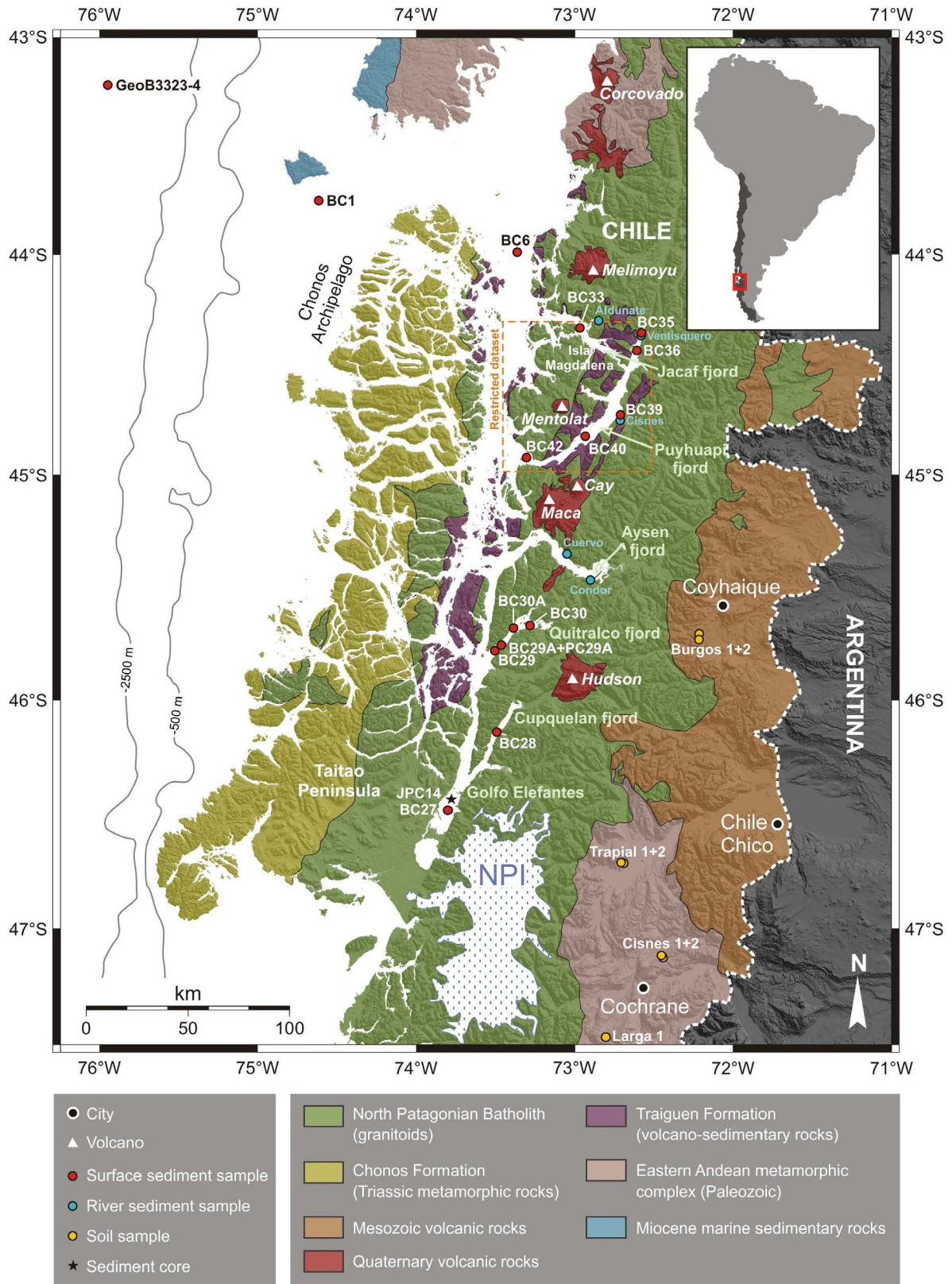
Site	Latitude	Longitude	Depth/ Elevation (m)	Salinity ^a (psu)
Fjord surface sediment samples				
BC1	-43.7517	-74.6215	-240	32.84
BC6	-43.9877	-73.3658	-176	31.18
BC27	-46.4843	-73.8042	-112	22.42
BC28	-46.1408	-73.4955	-239	23.33
BC29	-45.7812	-73.5087	-114	27.67
BC29A	-45.7560	-73.4673	-112	27.59
BC30	-45.6698	-73.2852	-269	27.98
BC30A	-45.6810	-73.3898	-110	27.92
BC33	-44.3335	-72.9695	-582	28.55
BC35	-44.3562	-72.5825	-52	22.24
BC36	-44.4368	-72.6110	-219	23.28
BC39	-44.7273	-72.7145	-160	22.66
BC40	-44.8247	-72.9345	-260	22.40
BC42	-44.9202	-73.3073	-320	28.93
River sediment samples				
Aldunate	-44.3000	-72.8500	0	
Ventisquero	-44.3676	-72.5833	0	
Cisnes	-44.7342	-72.7166	0	
Cuervo	-45.3500	-73.0508	0	
Condor	-45.4666	-72.9008	0	
Soil samples				
Burgos soil 1	-45.7131	-72.2174	+380	
Burgos soil 2	-45.7062	-72.2153	+390	
Trapial soil 1	-46.7126	-72.7110	+265	
Trapial soil 2	-46.7161	-72.6944	+250	
Cisnes soil 1	-47.1166	-72.4550	+435	
Cisnes soil 2	-47.1166	-72.4527	+435	
Larga soil 1	-47.4683	-72.8066	+270	
Open ocean surface sediment sample^b				
GeoB3323-4	-43.2183	-75.9500	-3697	

	Sampling station	Depth (m)	TI ^(a)	Salinity ^(b) (psu)	F _{terr} ^(c) (wt.%)	Bio-opal (wt.%)	Carbonate (wt.%)	TOC (wt.%)	lithogenic (wt.%)	Amphibole (wt.%)	Quartz (wt.%)	GS ^(d) (µm)	MS ^(e) (10 ⁻⁶ SI)	Fe in litho. ^(f) (wt.%)	Al in litho. ^(g) (wt.%)	CIA ^(h)	CIW ⁽ⁱ⁾
Fjord surface sediment samples (n=14)	1	240	-2.44	32.84	38.12	3.89	2.14	1.00	91.77	7	16	82.48	6824	4.87	8.61	46.58	48.57
	6	176	-1.97	31.18	41.97	7.48	1.89	2.14	85.92	10	21	30.65	2500	5.30	8.59	47.05	46.48
	27	112	1.74	22.42	98.85	5.30	0.07	0.51	93.50	14	26	15.40	1407	6.20	8.77	49.64	50.15
	28	239	1.22	23.33	88.80	10.02	0.04	0.83	88.11	9	14	7.00	777	6.73	8.96	48.49	45.74
	29	114	-1.13	27.67	45.61	8.80	0.06	0.50	90.04	0	6	31.54	1216	5.35	8.95	51.52	45.67
	29A	112	-0.27	27.59	70.62	12.36	0.04	1.36	84.61	16	5	21.45	2093	6.02	8.95	47.84	43.89
	30	269	-0.12	27.98	77.49	17.56	0.37	3.03	75.41	0	9	17.29	747	6.55	8.00	52.06	40.44
	30A	110	-0.55	27.92	64.34	17.69	0.10	2.06	77.68	0	14	12.04	965	6.60	8.52	59.37	41.92
	33	582	-0.91	28.55	57.53	11.26	2.04	3.73	78.50	10	15	10.18	710	5.99	8.36	41.39	35.05
	35	52	1.28	22.24	84.00	5.86	0.20	2.20	89.09	29	8	19.32	1972	6.99	8.23	40.90	40.63
	36	219	1.22	23.28	88.76	4.38	0.32	1.34	92.34	19	20	30.54	2599	6.23	8.62	46.00	45.03
	39	160	1.73	22.66	100.00	1.86	0.05	2.74	92.05	6	30	55.45	3061	4.18	8.02	51.48	51.06
	40	260	1.51	22.40	91.80	6.92	0.17	2.30	87.83	4	10	43.39	2615	6.01	9.10	47.67	41.77
	42	320	-1.31	28.93	47.97	12.18	2.34	2.87	79.15	6	11	18.24	1552	5.59	8.66	43.92	39.65
	mean	212	0.00	26.36	71.13	8.97	0.70	1.90	86.14	9	15	28.21	2074	5.90	8.60	48.14	44.00
	s.d. ⁽ⁱ⁾	132	1.44	3.57	21.87	4.84	0.93	1.00	6.11	8	7	20.54	1576	0.77	0.35	4.73	4.42
	Avg river sed. (n=5)						0.01	1.69	94.80	30	22	80.78	16074	7.00	8.56	46.44	48.22
	s.d. ⁽ⁱ⁾						0.00	2.26	6.96	12	15	7.48	9403	1.30	0.62	2.49	3.30
	Avg soil sed. (n=7)						0.05	4.18	89.14	0	32	38.73	2214	6.79	10.45	66.73	70.95
	s.d. ⁽ⁱ⁾						0.04	1.88	6.33	0	25	8.48	1071	1.44	1.50	6.03	7.24

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Bertrand et al – Table 2



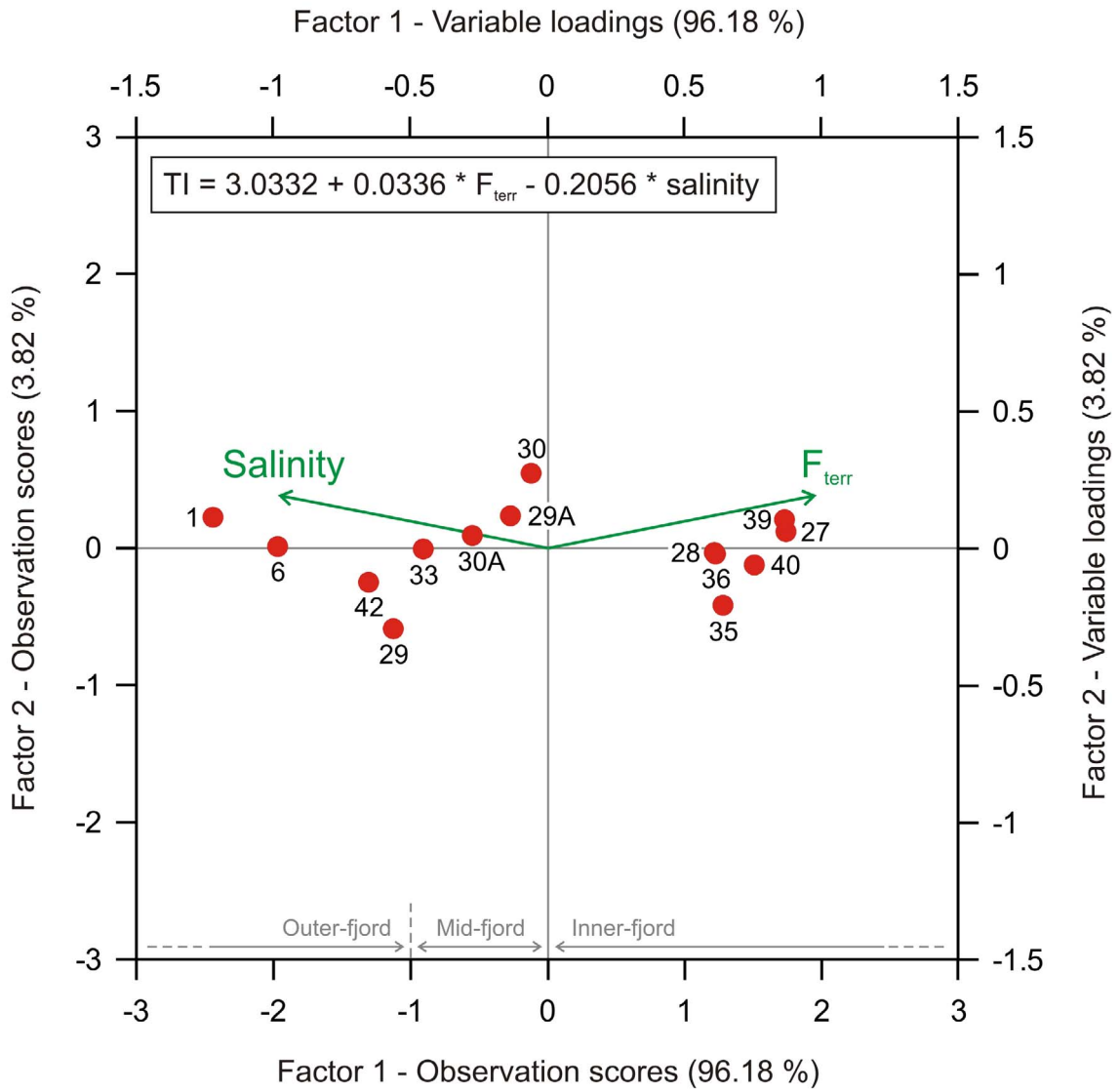
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Bertrand et al - Figure 1

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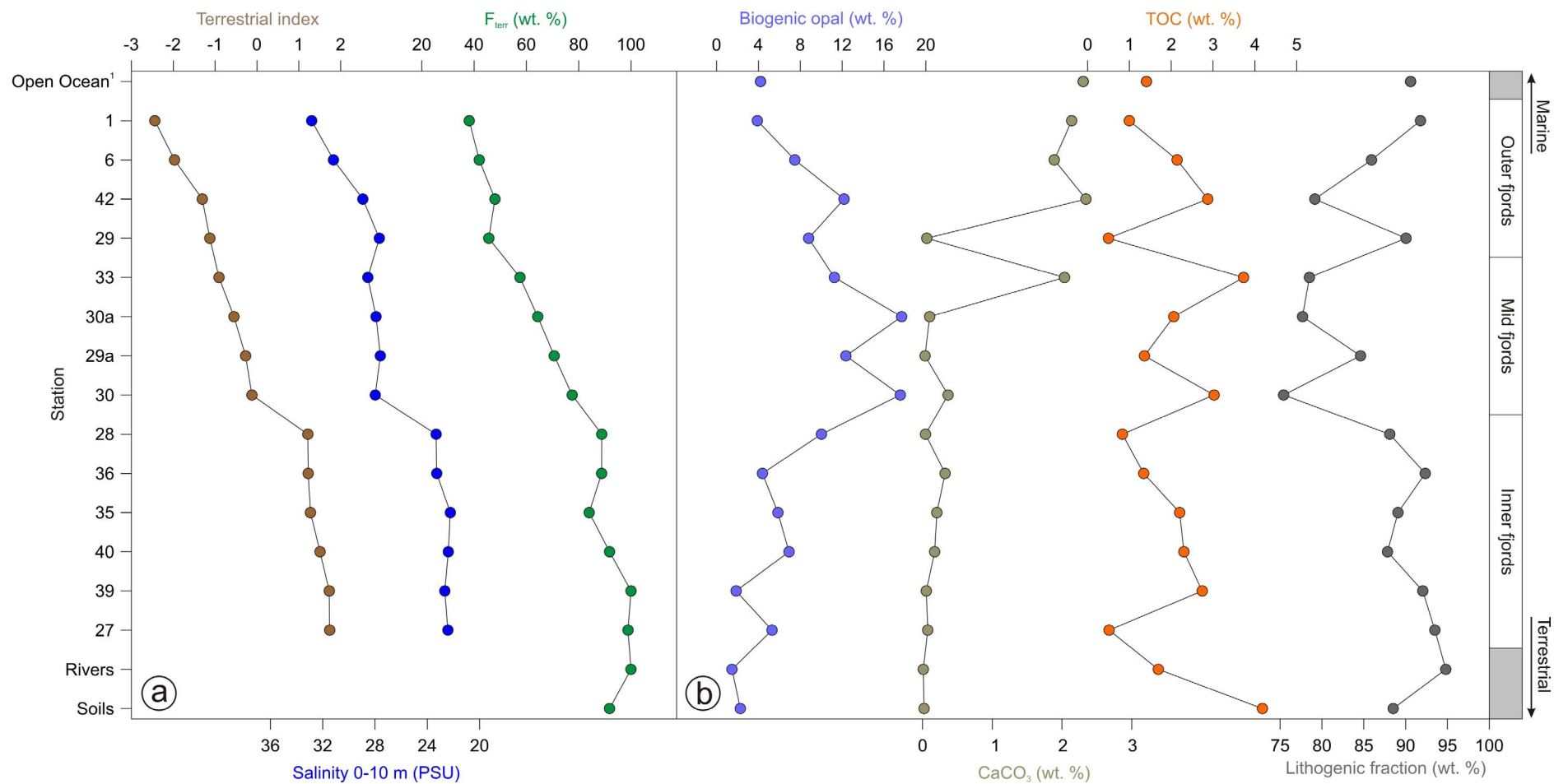
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Bertrand et al - Figure 2



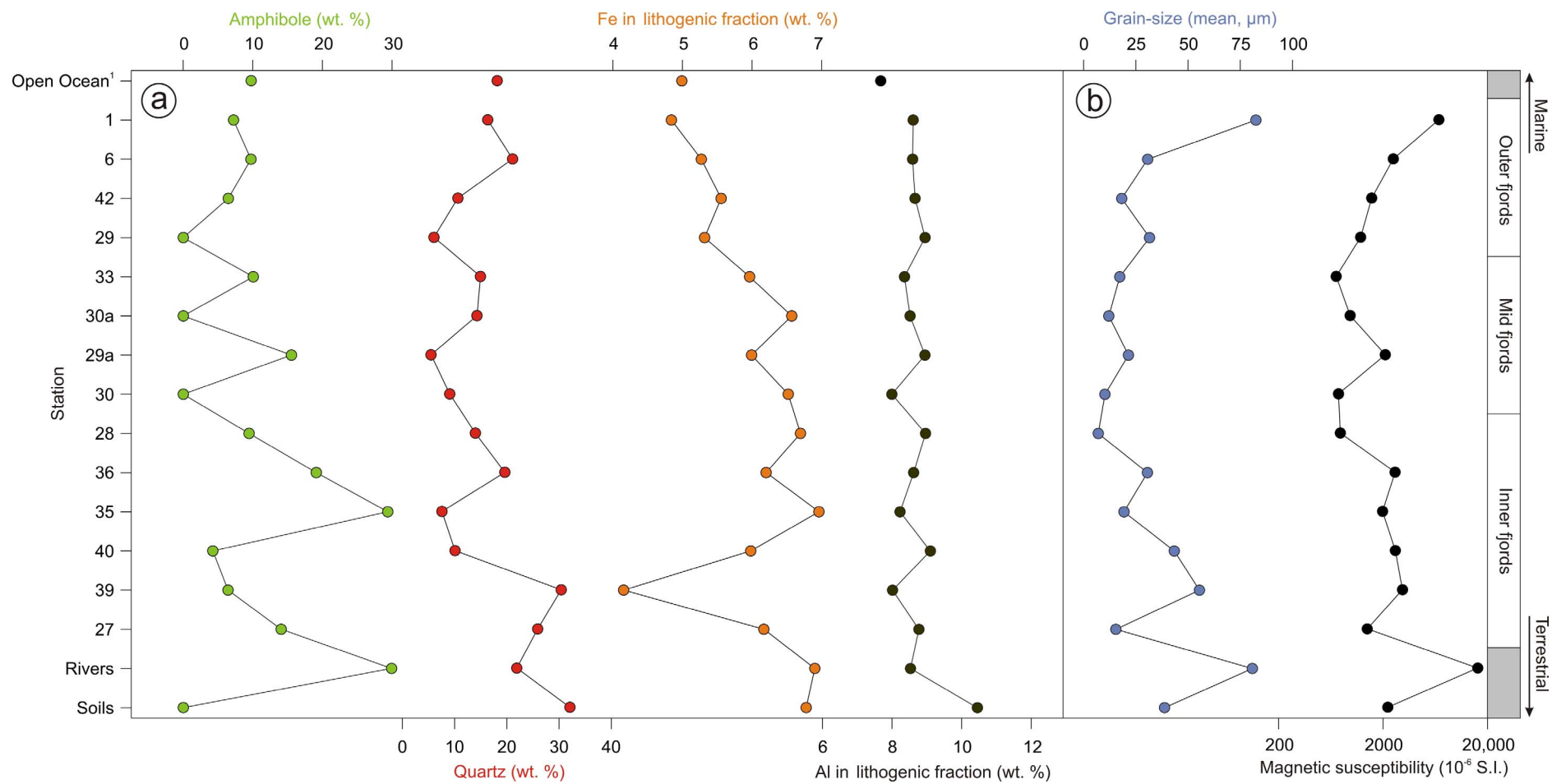
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Bertrand et al - Figure 3



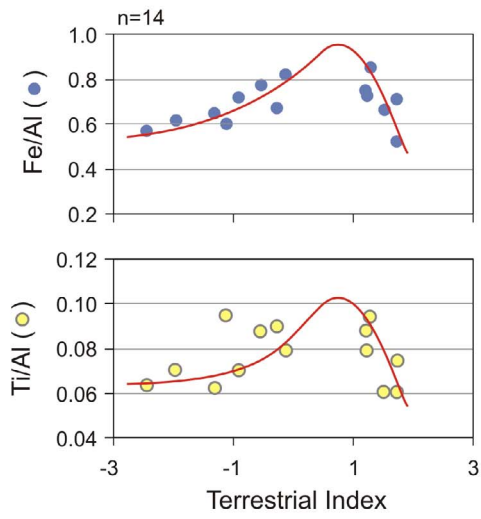
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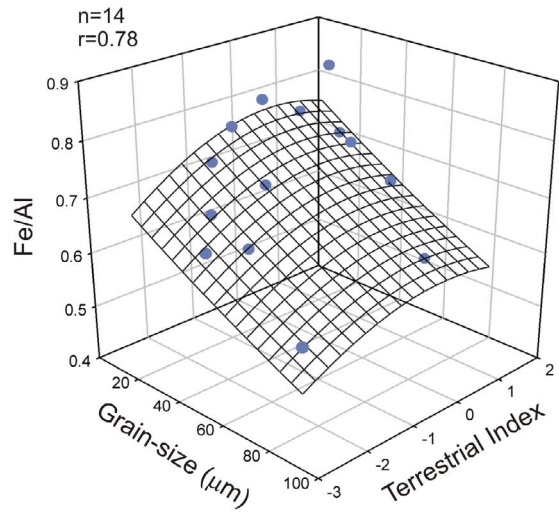
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Bertrand et al – Figure 4

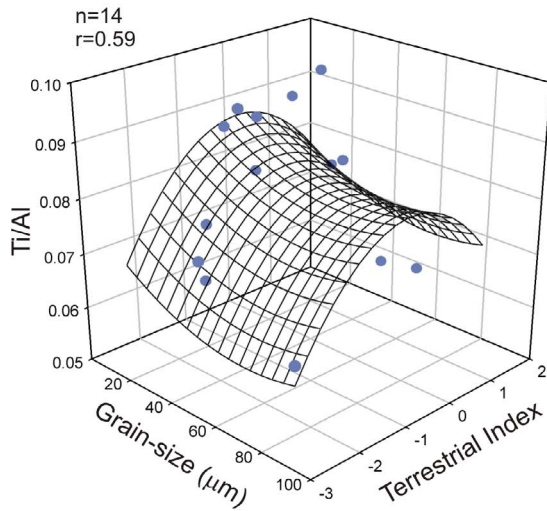
(a) Fe/Al, Ti/Al and Terrestrial Index



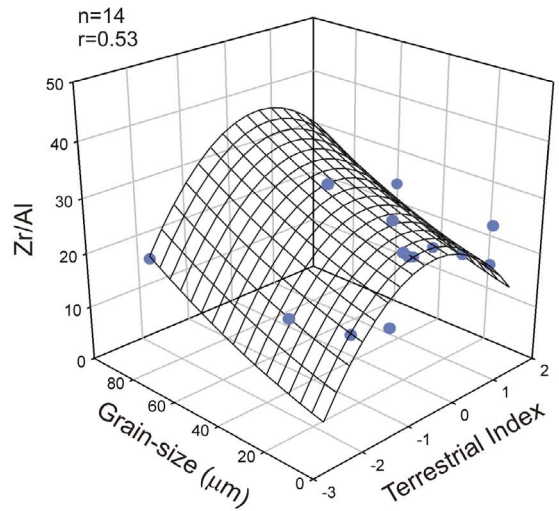
(b) Fe/Al, grain-size and Terrestrial Index



(c) Ti/Al, grain-size and Terrestrial Index



(d) Zr/Al, grain-size and Terrestrial Index



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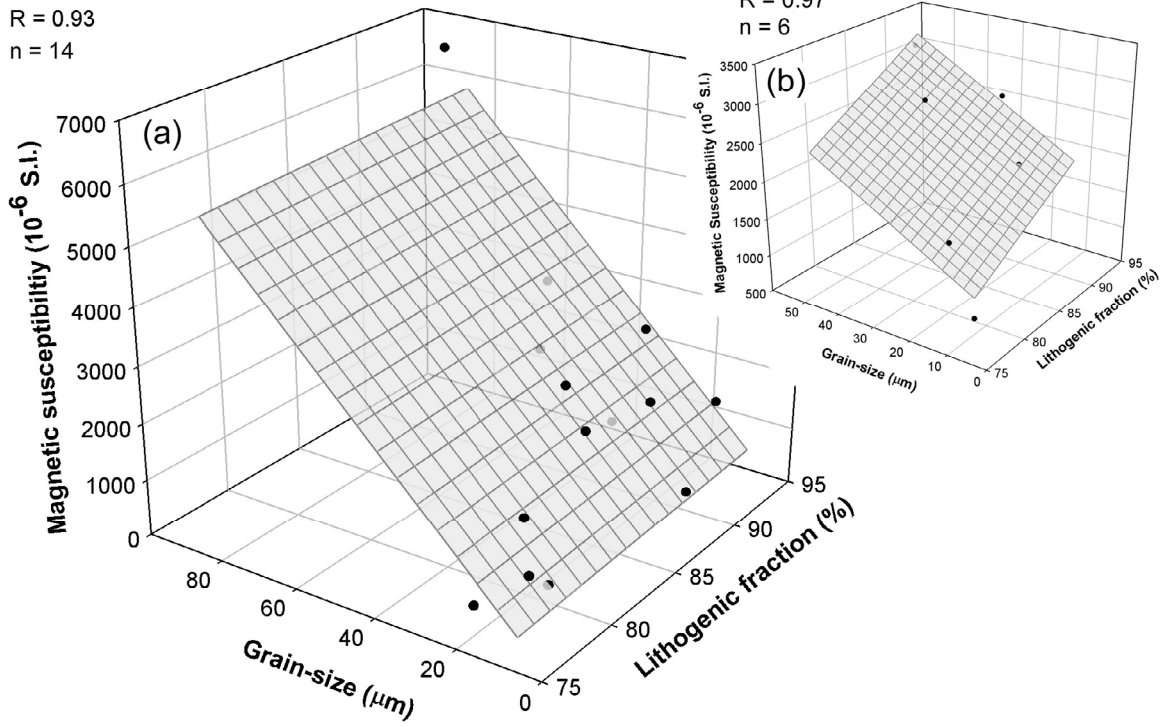
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Bertrand et al - Figure 5

Magnetic susceptibility, grain-size and lithogenic fraction

MS = 11.29 terr + 69.65 GS - 864.11
R = 0.93
n = 14



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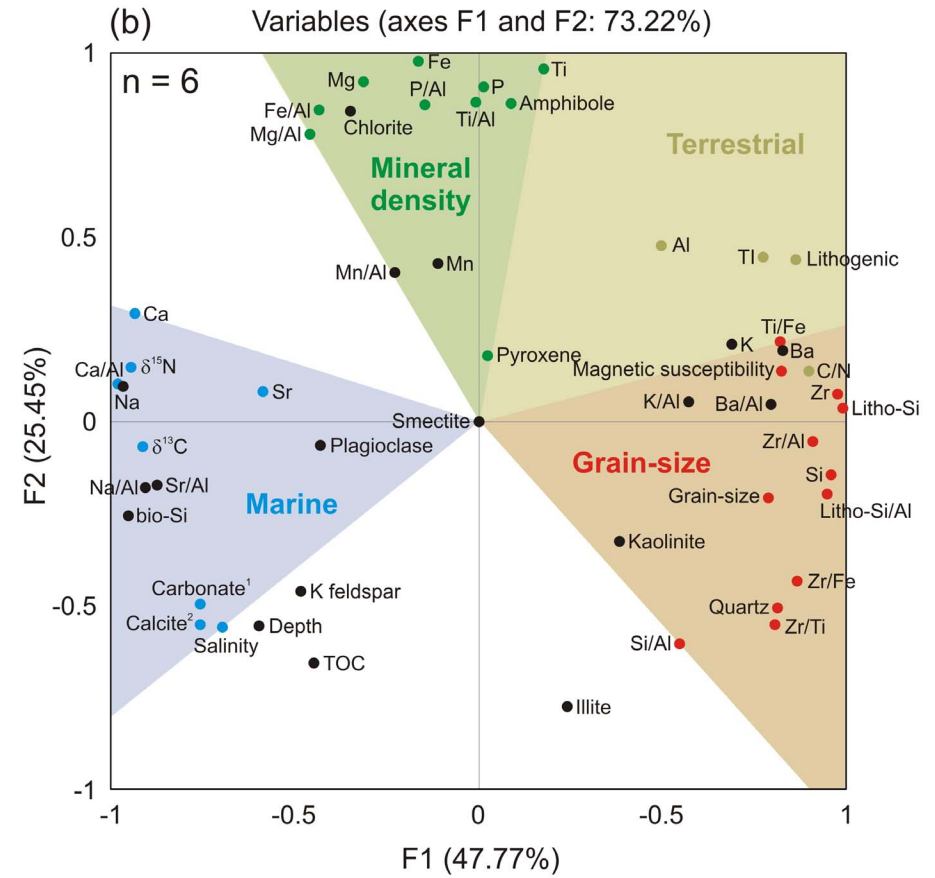
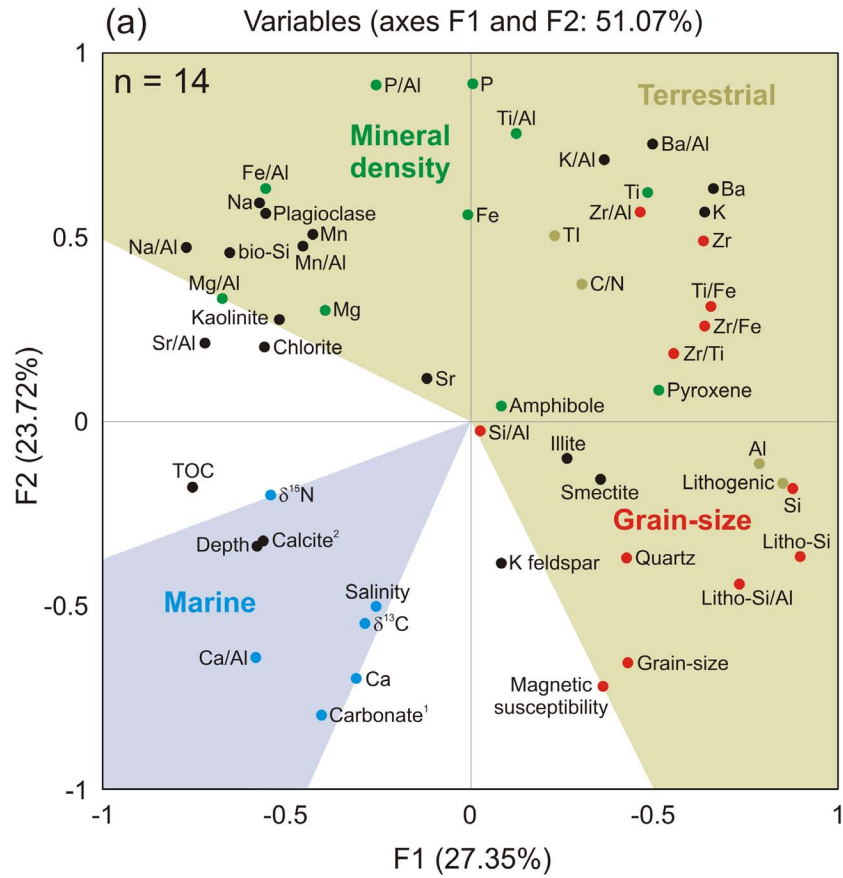
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Bertrand et al – Figure 6

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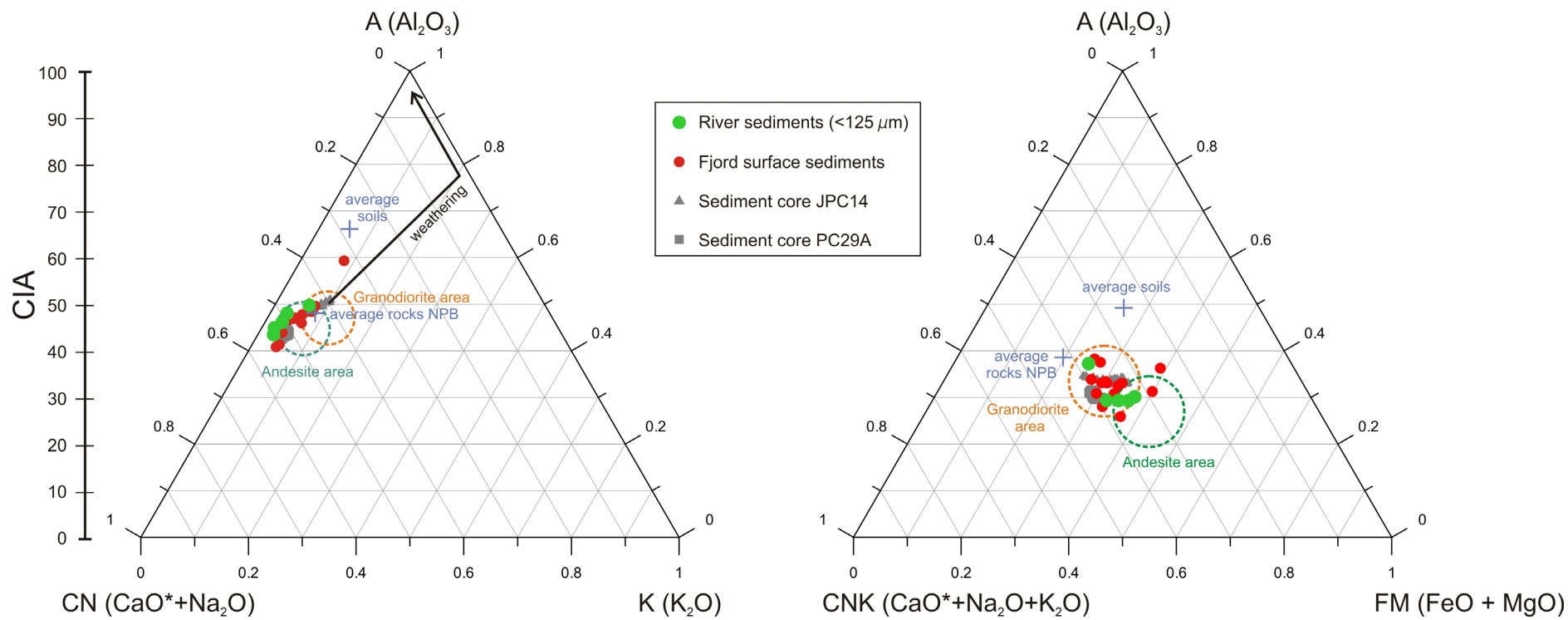


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Bertrand et al – Figure 7



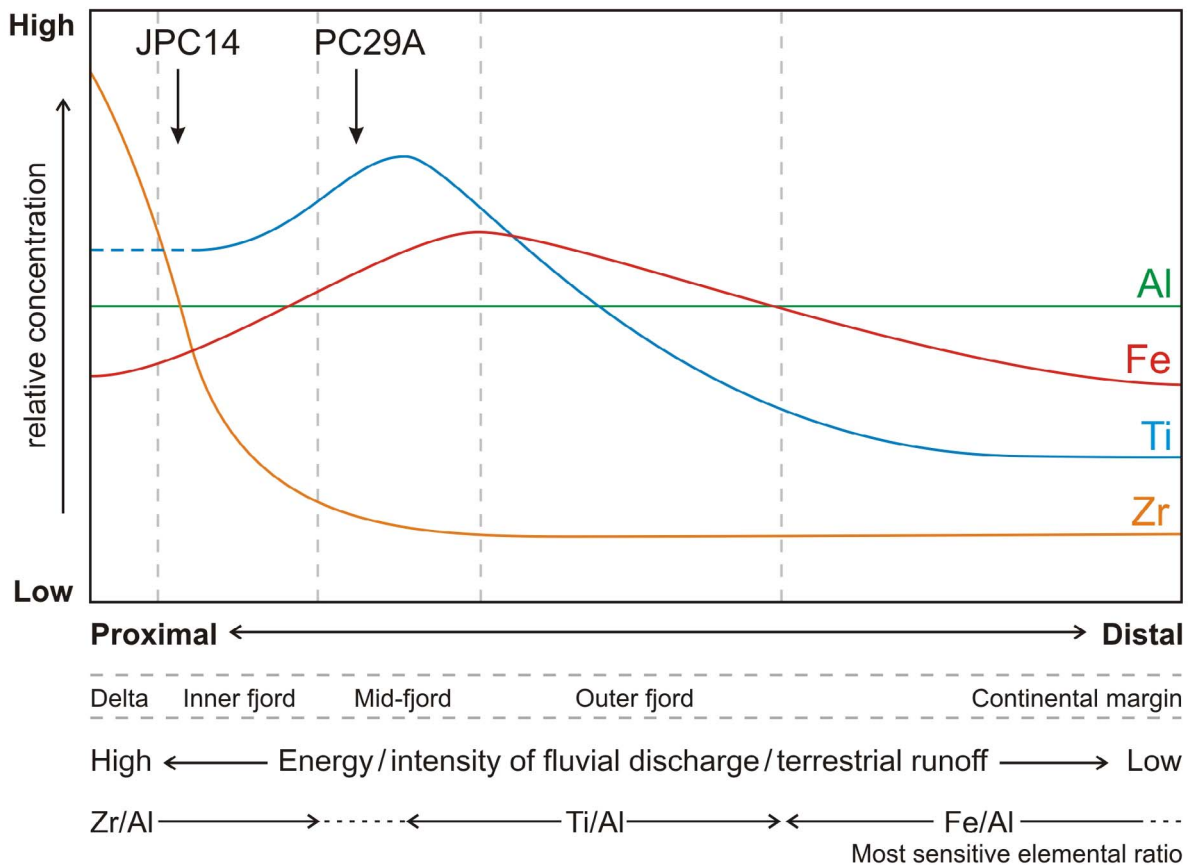
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Bertrand et al - Figure 8

Concentration in lithogenic fraction

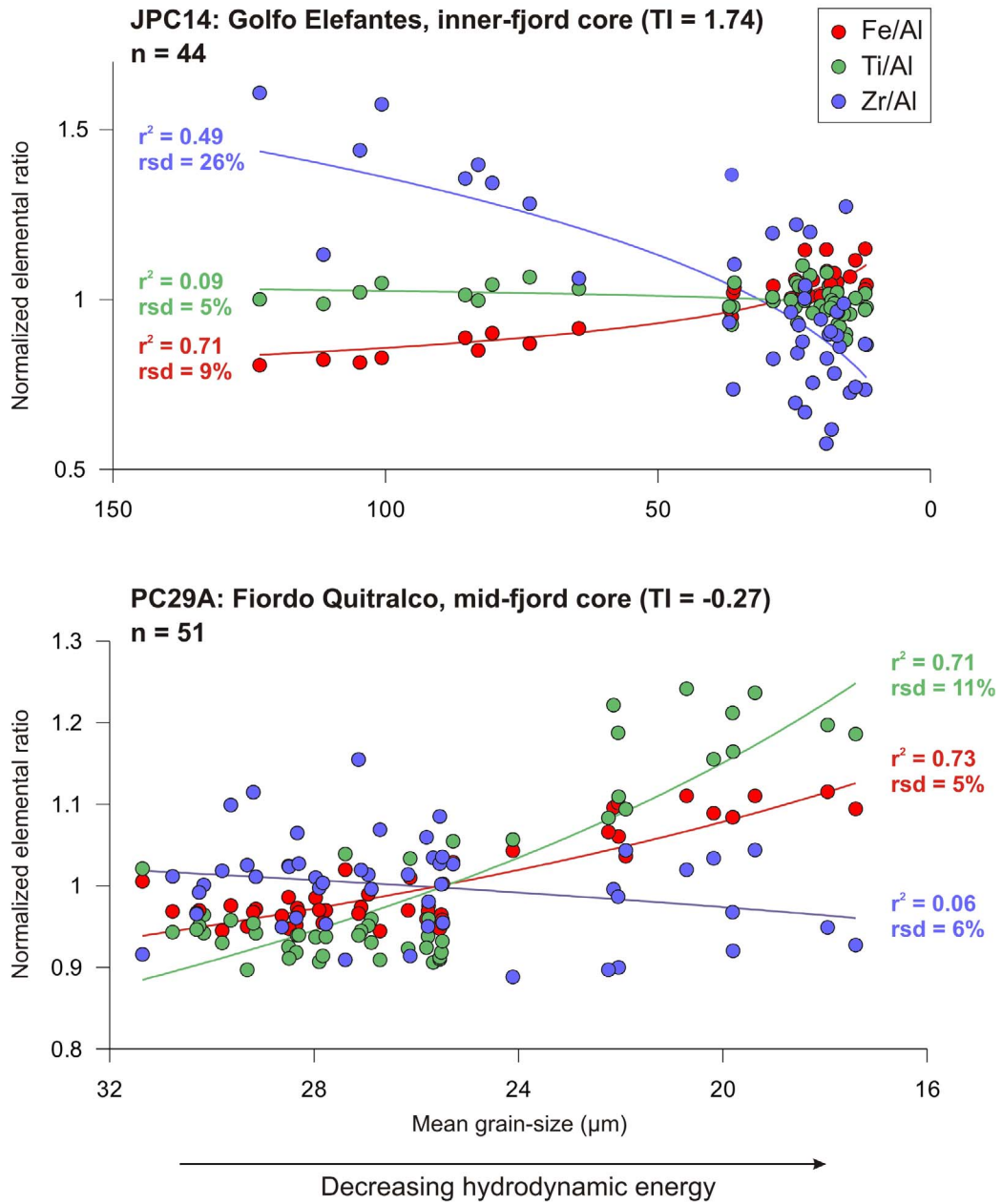


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Bertrand et al - Figure 9



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Bertrand et al - Figure 10