

1 **Supplementary Material**

2

3 *Experimental*

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5 Most data are presented here were previously published in the literature or from new analysis
6 of some of the natural gases (*Table 1*). In the case of the Icelandic DICE samples for which
7 no gas composition was available, we analyzed gases extracted from aliquots of DICE11 and
8 AO12, another glass sampled at the same site, by vacuum crushing, manometric
9 measurement, and mass spectrometry. Between 0.5 and 1 g of fresh glass devoid of alteration
10 features and washed in acetone were loaded in crushing tubes and baked overnight at 100°C.
11 The cm-sized glass fragments were crushed by activating with external solenoids an iron
12 piston (Marty and Zimmermann, 1999). The total pressure of extracted gas was measured
13 using a Baratron[®] in a calibrated volume. The gas was then purified and the He amount and
14 $^3\text{He}/^4\text{He}$ ratio were measured. The analysis of a gas aliquot by gas chromatography showed
15 CO_2 to be the major species (>90 %). Results are given in Table S1. For the DICE11 sample,
16 the amount of ^4He ($0.5\text{-}1.3 \times 10^{-10}$ mol/g for different aliquots) is comparable to the total
17 amount of ^4He recovered from sample DICE10 ($1.2\pm 0.3 \times 10^{-10}$ mol/g) by Mukhopadhyay
18 (2012), the differences being probably due to vesicle density heterogeneities. Likewise, the
19 measured $^3\text{He}/^4\text{He}$ ratios of DICE11 (17.90 ± 0.35 Ra) and A012 (17.96 ± 0.58 Ra), and the
20 $\text{CO}_2/\beta\text{He}$ ratios of the two samples ($2.86\pm 0.28 \times 10^9$ and $3.17\pm 0.23 \times 10^9$, respectively) are
21 similar even if the amount of gas in AO12 is a factor of 2 higher than in DICE11. This
22 strongly suggests that gases in the respective samples originate from a common magma
23 source. Since we are interested in abundance ratios, heterogeneities in the total gas amounts
24 should not impact our conclusions. The nitrogen content, which could not be analyzed at
25 CRPG at present, was estimated from the Ar-N₂ analysis done previously by Marty and

26 Dauphas (2003). Repeated analysis of 5 aliquots showed an excellent homogeneity of the
27 $N_2/^{40}Ar^*$ ratio (72.5 ± 4.4 ; 1 STD) (Marty and Dauphas, 2003). We use the $^4He/^{40}Ar^*$ ratio of
28 2.26 ± 0.41 determined for DICE10 (Mukhopadhyay, 2012) as representative of the AO12 and
29 DICE samples to reconstruct the gas compositions given in *Table 1*.

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31 **Computing the C/N and $^{36}Ar/N$ of the BSE**

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33 In order to compute the C/N and $^{36}Ar/N$ of the BSE, we carried out Monte Carlo simulations
34 (10^7 iterations) based on the C, N and ^{36}Ar elemental concentrations of the surface and mantle
35 reservoirs summarized in Table 4. Note that all concentrations used here are normalized to the
36 mass of the silicate Earth ($4 \cdot 10^{27}$ g) for comparison propose.

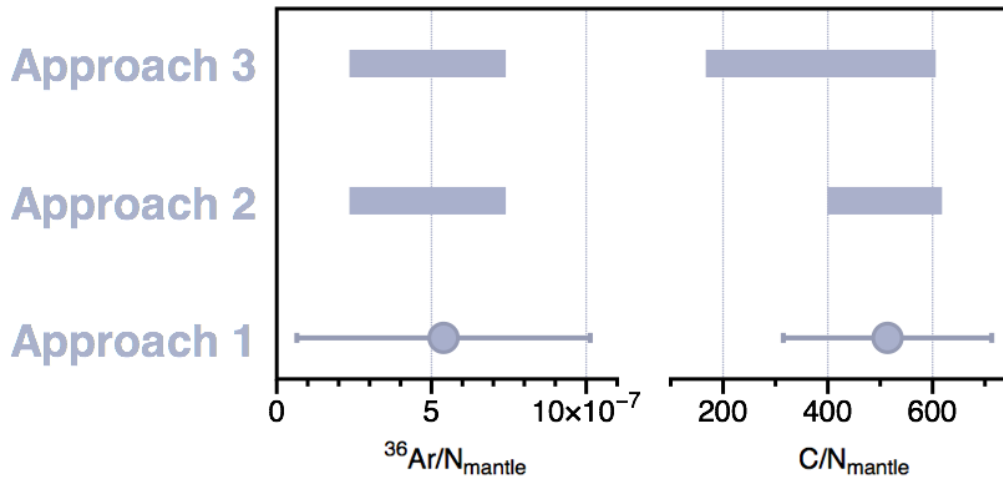
37

38 Basically, at each iteration step of the simulation, the concentrations of carbon ([C]) and
39 nitrogen ([N]) of the surface reservoir are generated randomly from normal distributions
40 corresponding to 1.56 ± 0.06 (ppmw) and 27 ± 5 (ppmw), respectively. Note that negative
41 concentrations are forbidden. The C/N of the surface is hence derived by computing the ratio
42 of [C] and [N]. The $^{36}Ar/N$ of the surface reservoir is then generated randomly from a normal
43 distribution corresponding to $(1.24 \pm 0.05) \cdot 10^{-5}$, therefore allowing the ^{36}Ar concentration
44 $[^{36}Ar]$ of the surface reservoir to be computed as $[^{36}Ar] = ^{36}Ar/N * [N]$.

45

46 Regarding the mantle, the [N] is generated randomly from normal distributions corresponding
47 to 1.10 ± 0.55 (ppmw). In order to compute [C] and $[^{36}Ar]$, we need to compute the C/N and
48 $^{36}Ar/N$ of the mantle, which can be defined following three different approaches.

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51

52 **Fig. S1:** $^{36}\text{Ar}/\text{N}$ and C/N values used for Monte Carlo simulations in our three approaches.

53 Approach 1 only considers the magmatic gas data and uses means and standard deviations to

54 compute $^{36}\text{Ar}/\text{N}$ and C/N . *Approach 2* only considers the magmatic gas data and uses

55 interquartile ranges (IQR). *Approach 3* considers both MORB and magmatic gas data and also

56 uses interquartile ranges (IQR).

57

58 First of all, C/N derived from the analysis of magmatic gas yield consistent values that

59 defined a mean composition of 514 ± 199 (*Approach 1; Fig. S1*). The median (474) and

60 interquartile range ($\text{IQR}=\text{Q3}-\text{Q1}=618-399=219$) corresponding to C/N values from magmatic

61 gas (*approach 2; Fig. S1*) are broadly consistent with the mean value (Fig. S1). Another way

62 of estimating the C/N of the mantle is to take into account all available values in the literature

63 (including MORB and magmatic gas samples). Given the disparities between C/N values

64 from MORB and magmatic gas samples, the mean C/N of the mantle derived by this method

65 would be 532 ± 599 , which cannot be used here for the purpose of our Monte Carlo approach.

66 The corresponding median value is 352, with an IQR of $\text{Q3}-\text{Q1}=606-167=439$ (*Approach 3;*

67 *Fig. S1*). Hence, for approach 1, the C/N values of the mantle were generated randomly from

68 normal distributions corresponding to 514 ± 199 (negative values also prohibited). For

69 approaches 2 and 3, the C/N values of the mantle were generated randomly from uniform
70 distributions within the ranges 399-618 and 167-606, respectively. The C concentrations of
71 the mantle were then computed as $[C] = C/N * [N]$. The $^{36}\text{Ar}/N$ values of the mantle were
72 only computed from magmatic gas data, defining a mean value of $(5.39 \pm 4.74) \cdot 10^{-7}$
73 (approach 1), and median and IQR of $2.90 \cdot 10^{-7}$ and $Q3-Q1 = 7.40 \cdot 10^{-7} - 2.35 \cdot 10^{-7} = 5.05 \cdot 10^{-7}$,
74 respectively (approach 2). For the sake of consistency, $^{36}\text{Ar}/N$ values were also generated
75 randomly from a uniform distribution within the range $2.35 \cdot 10^{-7} - 7.40 \cdot 10^{-7}$ for approach 3. The
76 ^{36}Ar concentration of the mantle was then computed as $[^{36}\text{Ar}] = ^{36}\text{Ar}/N * [N]$.
77
78 Lastly, the concentration of C in the BSE was computed by summing $[C]$ of the surface and of
79 the mantle. The concentration of C in the BSE was computed the same way. The C/N of the
80 BSE was finally computed by as $C/N = [C]_{\text{BSE}}/[N]_{\text{BSE}}$.

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82 **Approach 1:**

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84 C/N_{BSE} : mean = 218 ± 101

85 median = 210, 25th percentile (Q1) = 143, 75th percentile (Q3) = 283.

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87 $^{36}\text{Ar}/N_{\text{BSE}}$: mean = $7.73 \cdot 10^{-6}$

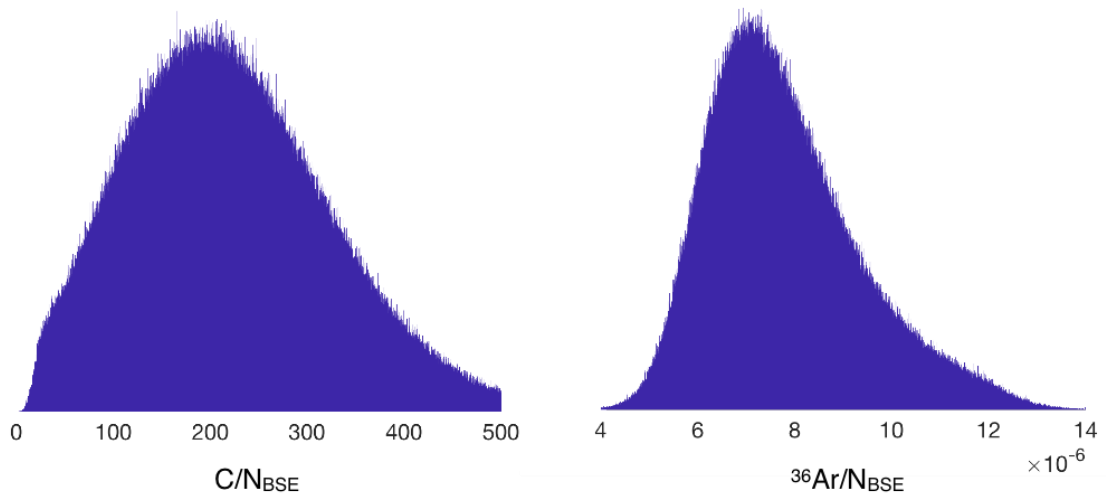
88 median = $7.51 \cdot 10^{-6}$, Q1 = $6.65 \cdot 10^{-6}$, Q3 = $8.59 \cdot 10^{-6}$.

89

90 $[C]_{\text{mantle}}$ (ppmw): median = 453, Q1 = 269, Q3 = 679.

91 $[Ar]_{\text{mantle}}$ (ppmw) : median = $1.54 \cdot 10^{-6}$, Q1 = $7.44 \cdot 10^{-7}$, Q3 = $2.66 \cdot 10^{-6}$.

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93

94 **Fig. S2:** Probability distributions of the C/N_{BSE} and $^{36}Ar/N_{BSE}$ obtained from approach 1.

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97 **Approach 2:**

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99 C/N_{BSE} : mean = 214

100 median = 219, 25th percentile (Q1) = 174, 75th percentile (Q3) = 259

101

102 $^{36}Ar/N_{BSE}$: mean = $7.66 \cdot 10^{-6}$

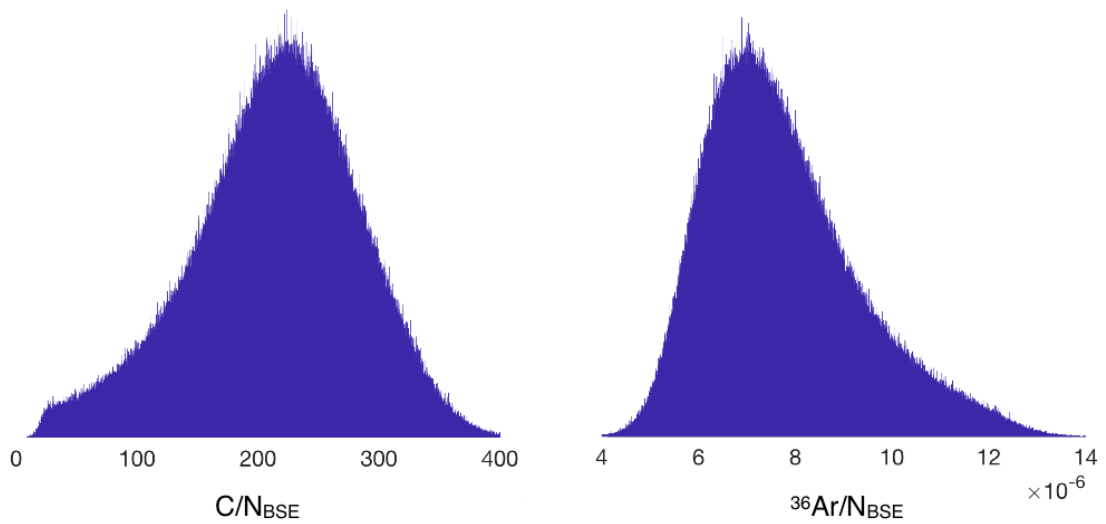
103 median = $7.43 \cdot 10^{-6}$, Q1 = $6.57 \cdot 10^{-6}$, Q3 = $8.52 \cdot 10^{-6}$

104

105 $[C]_{mantle}$ (ppmw): median = 479, Q1 = 323, Q3 = 646.

106 $[Ar]_{mantle}$ (ppmw) : median = $1.29 \cdot 10^{-6}$, Q1 = $8.24 \cdot 10^{-7}$, Q3 = $1.89 \cdot 10^{-6}$.

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108

109 **Fig. S3:** Probability distributions of the C/N_{BSE} and $^{36}Ar/N_{BSE}$ obtained from approach 2.

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111 **Approach 3:**

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113 C/N_{BSE} : mean = 166

114 median = 158, 25th percentile (Q1) = 112, 75th percentile (Q3) = 215

115

116 $^{36}Ar/N_{BSE}$: mean = $7.66 \cdot 10^{-6}$

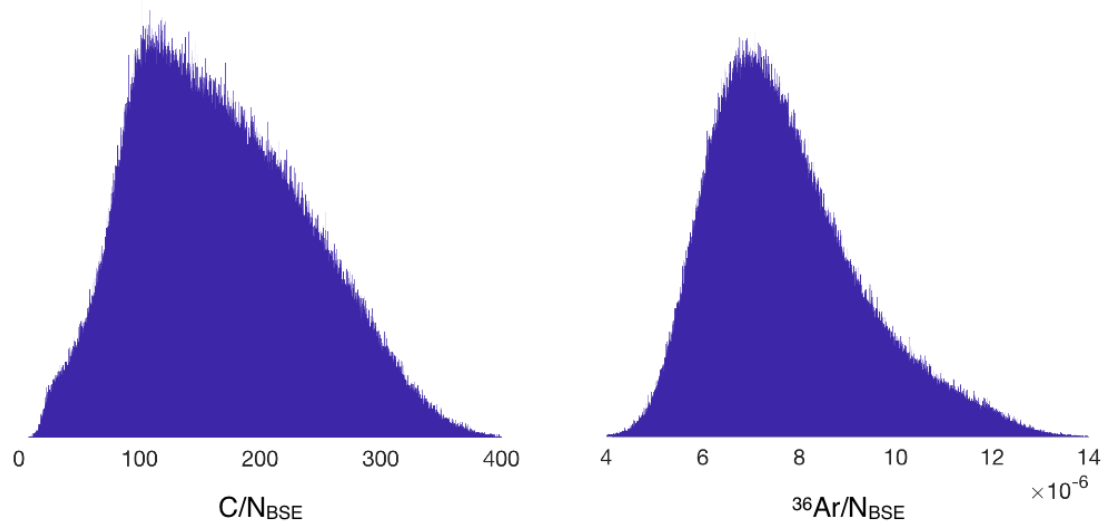
117 median = $7.43 \cdot 10^{-6}$, Q1 = $6.57 \cdot 10^{-6}$, Q3 = $8.52 \cdot 10^{-6}$

118

119 $[C]_{mantle}$ (ppmw): median = 337, Q1 = 211, Q3 = 504.

120 $[Ar]_{mantle}$ (ppmw) : median = $1.29 \cdot 10^{-6}$, Q1 = $8.24 \cdot 10^{-7}$, Q3 = $1.89 \cdot 10^{-6}$.

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124 **Fig. S4:** Probability distributions of the C/N_{BSE} and $^{36}Ar/N_{BSE}$ obtained from approach 3.

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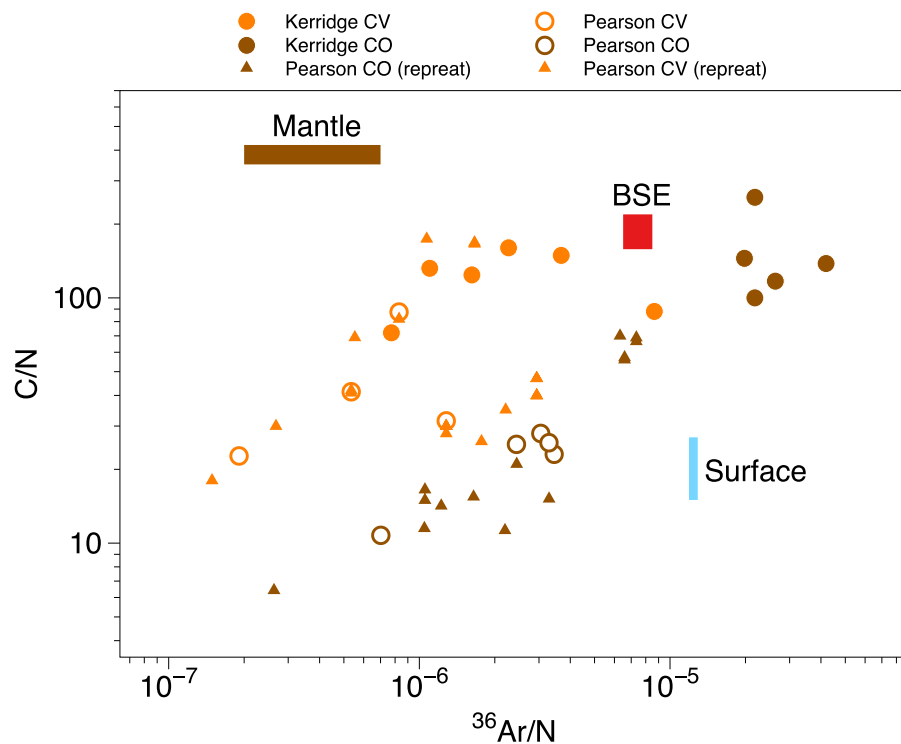
127 **Supplementary Table S1:** Compilation of meteorite C, N and ^{36}Ar data (attached).

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129

130 **Supplementary Figure S5 and discussion of meteorite data**

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132

133 **Fig. S5.** Nitrogen concentration data for CO and CV chondrites is scant in the literature. In
 134 the main text we use the values given by Kerridge, (1985), however we note that Pearson et
 135 al., (2006) also analyzed N within CO and CV chondrites. In this figure we plot the bulk C/N
 136 and $^{36}\text{Ar}/\text{N}$ for all CO and CV chondrites measured in these studies. The mean C/N and
 137 $^{36}\text{Ar}/\text{N}$ values for Pearson et al., (2006) are consistently lower than those calculated from the
 138 C and N concentrations given by Kerridge, (1985). One possibility for the discrepancy
 139 between the two studies is that the mass of sample analyzed by Pearson et al., (2006) is 1-2
 140 orders of magnitude lower than that used by Kerridge, (1985). This results in large
 141 uncertainties (up to 100%) on the N concentration within the CO and CV chondrites measured
 142 by Pearson et al., (2006). Another possibility that may explain the difference between these
 143 studies is that Kerridge, (1985) pre-heated the samples to 200°C prior to analysis. This may
 144 have reduced atmospheric contamination but could have also resulted in the preferential
 145 release of N relative to C and ^{36}Ar . We note however, that the C/N and $^{36}\text{Ar}/\text{N}$ of CM
 146 chondrites as calculated from Kerridge, (1985) is consistent with that calculated by Pearson et
 147 al., (2006), Alexander et al., (2012) and Alexander et al., (2013), who did not pre-heat their

148 samples (Supplementary Table 1). We therefore do not consider the preferential loss of N
149 during preheating to be a significant factor in explaining the difference between these studies.
150 Finally, we highlight that whilst the mean data from Pearson et al., (2006) is consistently
151 lower than that calculated from Kerridge, (1985), the individual analyses span a broader range
152 that approach the values of Kerridge, (1985). The small sample sizes used by Pearson et al.,
153 (2006) may therefore reflect the sample heterogeneity and we therefore consider the data from
154 Kerridge, (1985) to be better representative of the bulk chondrites (Fig. 5, main text).

155

156 **Sources of data presented in the meteorite compilation (*Table S1*)**

157

158 The names in the Table are those of the first author and the two following numbers refer to
159 the year of publication.

160

161 Alexander, C.M.O., Bowden, R., Fogel, M.L., Howard, K.T., Herd, C.D.K., Nittler, L.R.,
162 2012. The provenances of asteroids, and their contributions to the volatile inventories of
163 the terrestrial planets. *Science* 337, 721–3. doi:10.1126/science.1223474

164 Alexander, C.M.O., Howard, K.T., Bowden, R., Fogel, M.L., 2013. The classification of CM
165 and CR chondrites using bulk H, C and N abundances and isotopic compositions.
166 *Geochim. Cosmochim. Acta* 123, 244–260. doi:10.1016/j.gca.2013.05.019

167 Bischoff, A., Palme, H., Ash, R.D., Clayton, R.N., Schultz, L., Herpers, U., Stöffler, D.,
168 Grady, M.M., Pillinger, C.T., Spettel, B., Weber, H., Grund, T., Endreß, M., Weber, D.,
169 1993. Paired Renazzo-type (CR) carbonaceous chondrites from the Sahara. *Geochim.*
170 *Cosmochim. Acta* 57, 1587–1603. doi:10.1016/0016-7037(93)90014-N

171 Bogard, D.D., Clark, R.S., Keith, J.E., Reynolds, M.A., 1971. Noble gases and radionuclides
172 in Lost City and other recently fallen meteorites. *J. Geophys. Res.* 76, 4076–4083.

173 doi:10.1029/jb076i017p04076

174 Crabb, J., Anders, E., 1981. Noble gases in E-chondrites. *Geochim.* 45, 2443–2464.

175 Downes, H., Abernethy, F.A.J., Smith, C.L., Ross, A.J., Verchovsky, A.B., Grady, M.M.,
176 Jenniskens, P., Shaddad, M.H., 2015. Isotopic composition of carbon and nitrogen in
177 ureilitic fragments of the Almahata Sitta meteorite. *Meteorit. Planet. Sci.* 50, 255–272.
178 doi:10.1111/maps.12413

179 Eugster, O., Lorenzetti, S., Krähenbühl, U., Marti, K., 2007. Comparison of cosmic-ray
180 exposure ages and trapped noble gases in chondrule and matrix samples of ordinary,
181 enstatite, and carbonaceous chondrites. *Meteorit. Planet. Sci.* 42, 1351–1371.
182 doi:10.1111/j.1945-5100.2007.tb00579.x

183 Göbel, R., Ott, U., Begemann, F., 1978. On trapped noble gases in ureilites. *J. Geophys. Res.*
184 83, 855. doi:10.1029/JB083iB02p00855

185 Grady, M.M., Verchovsky, A.B., Franchi, I.A., Wright, I.P., Pillinger, C.T., 2002. Light
186 element geochemistry of the Tagish Lake CI2 chondrite : Comparison with CI1 and CM2
187 meteorites. *Meteorit. Planet. Sci.* 37, 713–735.

188 Grady, M.M., Wright, I.P., Carr, L.P., Pillinger, C.T., 1986. Compositional differences in
189 enstatite chondrites based on carbon and nitrogen stable isotope measurements.
190 *Geochim. Cosmochim. Acta* 50, 2799–2813.

191 Grady, M.M., Wright, I.P., Swart, P.K., Pillinger, C.T., 1985. The carbon and nitrogen
192 isotopic composition of ureilites : implications for their genesis 49.

193 Kerridge, F., 1985. Carbon, hydrogen and nitrogen in carbonaceous chondrites: abundances
194 and isotopic compositions in bulk samples. *Geochim. Cosmochim. Acta* 49, 1707–1714.

195 Mazor, E., Heymann, D., Anders, E., 1970. Noble gases in carbonaceous chondrites.
196 *Geochim. Cosmochim. Acta* 34, 781–824.

197 Murty, S.V.S., Mahajan, R.R., Jenniskens, P., Shaddad, M.H., Eldien, B., 2010. Noble gases

198 and nitrogen in the Almahata Sitta ureilite. *Meteorit. Planet. Sci.* 45, 1751–1764.
199 doi:10.1111/j.1945-5100.2010.01095.x

200 Nakamura, T., Noguchi, T., Zolensky, M.E., Tanaka, M., 2003. Mineralogy and noble-gas
201 signatures of the carbonate-rich lithology of the Tagish Lake carbonaceous chondrite:
202 evidence for an accretionary breccia. *Earth Planet. Sci. Lett.* 207, 83–101.
203 doi:10.1016/S0012-821X(02)01127-5

204 Okazaki, R., Nagao, K., 2017. Title Primordial and cosmogenic noble gases in the Sutter’s
205 Mill CM chondrite. *Meteorit. Planet. Sci.* 52, 669–689. doi:doi.org/10.1111/maps.12819

206 Ott U., Loehr H.P. and Begemann F. (1985) Trapped noble gases in 5 more ureilites and the
207 possible role of Q. *Lunar Planet. Sci.* **16**, 639-640.

208 Patzer, A., Schultz, L., 2001. Noble gases in enstatite chondrites I: Exposure ages, pairing,
209 and weathering effects. *Meteorit. Planet. Sci.* 36, 947–961. doi:10.1111/j.1945-
210 5100.2001.tb01932.x

211 Pearson, V.K., Sephton, M.A., Franchi, I.A., Gibson, J.M., Gilmour, I., 2006. Carbon and
212 nitrogen in carbonaceous chondrites: Elemental abundances and stable isotopic
213 compositions. *Meteorit. Planet. Sci.* 41, 1899–191