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INTRODUCTION

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concentrations. Agriculture and industry have strongly altered the N cycle in ways that impact the environment from local to global scales by contributing to increasing greenhouse gas emissions, acidic deposition, and impairing the functioning of ecosystems through the eutrophication of soils and waters. The N cycle is thus intrinsically coupled with various environmental processes and factors including the transformation of land-use, energy and food production and consumption, climate change, exploitation of natural resources, air, soil and water pollution, human health, ecosystem services, and other natural and anthropogenic drivers (Galloway et al. 2004). Future sustainable management and stewardship of both less disturbed and highly managed ecosystems require a more integrated approach to the assessment of linkages between these systems and their interaction with the human society (Collins et al. 2011). Long-term monitoring of N biogeochemistry is a powerful research approach to understanding the dynamic features of ecosystem behavior influenced by natural and anthropogenic drivers, locally, regionally and globally. The Long-Term Ecological Research (LTER) program was first established in early 1980s in the USA. Since then, the LTER has been expanded to many other countries as an integrated ecological research network that enables long-term site-based research, field experiments, and database development. The US-LTER has produced important findings on the N dynamics in a broad geographical range of watersheds and landscapes, including the long-term impact of atmospheric N deposition on forest ecosystems, the impact of logging on stream N chemistry, the climate impacts on N cycles in ecosystems, effects on biodiversity, etc. (e.g., Likens et al. 1996; Clark and Tilman 2008; Fernandez et al. 2010; Groffman et al. 2012; Driscoll et al. 2012). Some European countries have also conducted long-term environmental monitoring, for example Sweden (e.g., Löfgren, et al. 2011), the UK (Curtis et al., 2014) and Finland (Rask et al. 2014). In addition to the LTER sites, long-term N

Nitrogen (N) is an essential nutrient, but reactive N has well-known deleterious effects in high

experiments have been also been conducted at other sites in the USA and Europe. Examples of European studies include the NITREX (Gundersen et al. 1998) and climate experiments linked with N dynamics (e.g., CLIMEX (Wright et al. 1998), CLIMOOR (Beier et al. 2004), and VULCAN (Peñuelas et al. 2007)). Lake manipulation experiments have been undertaken in Canada, Norway, the USA and Finland (e.g., Carpenter et al 2001; Harris et al. 2014). There have also been a number of snow manipulation experiments in the USA, Norway and Germany (e.g., Kaste et al. 2008; Wipf and Rixen 2010). While contributing to our current understanding of N cycling in ecosystems, they also reveal significant gaps in knowledge which will require a continued commitment to long-term research and a broadening of international perspectives to address. The LTER program has expanded internationally since the early 1990s and the International LTER (ILTER) was created in 1993. This currently comprises over 600 sites within 40 member networks, providing great potential for understanding altered N biogeochemistry and its impact in different environment and socio-ecological settings at an international scale. However, effective integration of this international effort has been impeded by the absence of a mechanism to bring the international research community together and the relative paucity of robust and directly comparable data that could be employed in meta-analyses at global and regional scales. Recent syntheses of nitrogen issues in the USA and Europe (e.g. Sutton et al. 2011; Suddick and Davidson 2012) bring to the fore the need to review the contributions that long term environmental research has made to the scientific understanding of the changing N cycle and consider how it might address the current major gaps in knowledge. Here we review current understanding of the impact of anthropogenic N on various ecosystems and environments to elucidate the consequences of globally increased N cycles for coupled social-ecological systems under a changing climate (Fig. 1). Particular attention is paid to the most recent trends in anthropogenic reactive N emissions, including: nitrous oxide (N2O); the complex interaction of N with carbon (C), phosphorus (P) and other elements; impacts of N on biodiversity; seasonal and long-term trends in N biogeochemistry associated with climate variability; the N cascade process from

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49 terrestrial to aquatic ecosystem; and finally, societal challenges from ecosystem services to human health.

Emerging uncertainties and further research questions are also discussed.

INCREASED ANTHROPOGENIC NITROGEN EMISSIONS

Humans create more reactive N (Nr) than natural ecosystems do (Galloway et al. 2003), principally nitrogen oxides (NO_x -N, i.e., sum of N_2O , NO, and NO_2) and ammonia (NH_3), mostly through food and energy production and consumption and their various byproducts. Reactive N emissions into the atmosphere contribute to increasing greenhouse gases, acidic deposition, as well as excess inputs of N nutrients to receiving environments. One of the Nr forms, N_2O , is an important greenhouse gas with an exceptionally long atmospheric half-life that is emitted through agricultural activities as well as natural processes such as denitrification within wetlands. Here we review the long-term trends of Nr and N_2O emissions and their current knowledge gaps.

Global long-term trends in anthropogenic emissions of reactive nitrogen

Reactive N emissions associated with human-induced burning of biomass and animal husbandry have been entering the atmosphere for over 10,000 years. Emissions have increased sharply since the onset of the industrial revolution due to fossil fuel combustion, while the use of synthetic N-fertilizers became especially important in the 20th century. The global cumulative anthropogenic release of Nr to the atmosphere over the last 10,000 years has been estimated at ~17.4 Pg N, 28% of which was emitted during 1850–2000 and 42% during 1–1850 AD (Kopáček and Posch, 2011). Recent global emissions of NO_x from anthropogenic and natural sources have been estimated to range from 44 to 50 Tg N yr⁻¹, while the contribution from NH₃ has been estimated at 54 Tg N yr⁻¹ (Kopáček and Posch, 2011). About 70% of

global NH_3 emissions are closely related to food production and agricultural systems, predominantly livestock production and the use of synthetic N-fertilizers (Kopáček and Posch, 2011). Global Nr emissions have increased sharply since the 1950s (Fig. 2). Overall, Europe is the only continent where Nr emissions have begun to decrease in recent years (i.e. since the late 1980s) (Fig. 2). This decrease is attributed to NO_x emission controls on energy production, lower NH_3 emissions due to reductions in cattle production, and reduced use of synthetic N-fertilizer (Kopáček and Posch, 2011).

Greenhouse gas emissions

et al. 2012).

than CO₂ on a per molecule basis. The production of N₂O occurs during both denitrification and nitrification. N₂O in the atmosphere is estimated to have increased by 18% from its pre-industrial level (IPCC 2007). The rise is attributed primarily to human activities, particularly from agriculture and land use change.

Information on the processes influencing N₂O emissions from soils is sparse, particularly that on the roles of temperature, moisture, redox potential, pH, and substrate availability (Wallenstein et al. 2006). While agricultural soils are considered a major source of N₂O, the effect of N fertilizer on soil N₂O emissions remains highly uncertain (Davidson 2009; Zaehle et al. 2011). Emission factors (N₂O emissions per unit N addition) have been reported to vary between 0.1–7% of the N applied (Skiba and Smith 2000). Since multiple processes and drivers are involved, N₂O emissions are highly variable and often associated with "hotspots" (high emissions from small areas) and "hot moments" (high emissions for brief periods), making measuring, modeling, and up-scaling challenging (Groffman et al. 2009; Reay

Nitrous oxide is a potent greenhouse gas with a global warming potential that is ~300 times greater

High variability in the response of N_2O to N inputs indicates nonlinearity of the response function (Hoben et al. 2011). However, there is also large global variation in levels of N-inputs to agricultural systems, ranging from 588 kg N ha⁻¹ yr⁻¹ in a wheat-maize double cropping system in North China to 7 kg N ha⁻¹ yr⁻¹ in maize systems in western Kenya, resulting in very large uncertainty over global N_2O emissions (Vitousek et al. 2009). Assessing a decade of measurements from an ILTER site in Michigan, USA, Robertson et al. (2000) reported that N_2O fluxes were similar among different cropping systems, suggesting that N_2O fluxes were driven by soil N availability, rather than by additional N inputs. In contrast, Van Groenigen et al. (2010) found that N inputs stimulated a dramatic increase in N_2O emissions when fertilizer rates reached 301 kg N ha⁻¹ yr⁻¹, while N_2O emissions were small when fertilizer rates were ~180–190 kg N ha⁻¹ yr⁻¹ or lower. Given the paucity of data for areas with low rates of fertilization such as Africa, there is currently insufficient information for establishing the response function of N_2O fluxes following addition of N inputs in these systems (Van Groenigen et al. 2010).

Clearly, there is a need for considerable clarification of the factors determining N_2O emissions from N inputs at representative sites around the globe. For example, assessments are urgently required to determine the global impact of the African Green Revolution, called for by the former United Nation secretary general Kofi Annan. Agricultural productivity across Sub-Saharan Africa is expected to increase substantially by major increases in fertilizer use (up to $100 \text{ kg ha}^{-1} \text{ yr}^{-1}$, Sanchez et al. 2007). To assess the unintended N_2O emissions and N leaching to the local environments, more research is urgently needed to understand the N fluxes in response to inputs.

ALTERED NITROGEN BIOGEOCHEMISTRY IN ECOSYSTEMS

Increasing anthropogenic Nr emissions are a significant source of atmospheric N deposition to land and sea (Galloway et al. 2004) and also enhance global warming through N_2O emissions (IPCC 2007),

disturbing N pools, cycling and transport in and among ecosystems. In this section, we review the current understanding of the impact of increased N deposition on N biogeochemistry over a period of changing climate in various ecosystems, with special attention to ILTER's findings.

Long-term effects of N deposition in watershed N cycles and leaching

Excess atmospheric N deposition beyond the N requirement of the biota often causes N saturation, which has been observed in many forest ecosystems of Europe and the USA (Dise and Wright 1995; Aber et al. 1998). Elevated concentrations of NO₃⁻ in surface waters derived from non-point or point sources of N pollution usually indicate that there has been sufficient anthropogenic deposition of N for catchment soils to have reached a degree of N saturation (Stoddard 1994).

In recent years, attempts in the USA and Europe to reduce atmospheric emissions of acidic precursors and other pollutants have resulted in widespread reductions in sulfur (S) deposition, with corresponding substantial reductions in sulfate concentrations in run-off. However, in Western Europe, there have been less linear changes of atmospheric deposition with respect to N. While reductions in emissions of N have led to broadly comparable reductions in ammonium (NH₄⁺) deposition, reductions in NO₃⁻ deposition have been much lower (Fowler et al. 2007). The combination of uncertainties associated with these non-linearities between reduction of NH₄⁺ and NO₃⁻ and the various responses of N leaching to variation in winter climate make the future prediction of NO₃⁻ leaching to surface waters and the extent to recovery from acidification very difficult.

In contrast, further increases in N emissions in the East Asia region during the next few decades have been predicted as a consequence of rapid industrialization, urbanization and economic growth (e.g. Galloway et al. 2003, 2004; Fang et al. 2011). Fang et al. (2011) indicated that N deposition in China ranged 2.6 to 48 kg N ha⁻¹ yr⁻¹, while the threshold of N deposition to N saturation in Chinese forest

differs from that in the USA and Europe (i.e., the relatively high N leaching in some Chinese forests receiving low N input). Niu et al. (2010) reported that an experimental addition of N deposition (100 kg N ha⁻¹ yr⁻¹) for four years enhanced the ecosystem productivity by 27% in a temperate steppe ecosystem in China. Several comparative studies (Park et al. 2003; Fang et al. 2011; Mitchell 2011) suggest that regional climate, geology and hydrology result in different patterns and responses to elevated N deposition in forest ecosystems, when compared to previous findings of similar studies in the USA and Europe (Ohte et al. 2001), indicating that more comparative research is needed to generalize the impact of increased N deposition on ecosystems. Long-term studies of watershed N biogeochemistry are clearly limited outside the USA and Europe, especially in East Asia, South America and Africa, where anthropogenic N deposition will increase in future.

Complex interactions of N with other elements

The N cycle is intimately coupled to the C cycle. Soils with large pools of organic C and high C:N ratios are generally associated with N accumulation and tend to export less NO₃⁻ than soils with low C:N ratios (e.g., Aber et al. 1998; Gundersen et al. 1998). When Nr availability is elevated in an N-limited system (e.g. through atmospheric N deposition), soil inorganic N is readily utilized by plants, resulting in increased C uptake (Gruber and Galloway 2008) and reduced below-ground allocation of C (Deegan et al. 2012). Elevated Nr in soil can also change the soil microbial community by, for example, a reduction of fungal:bacterial biomass ratios (Högberg et al. 2007; Boberg et al. 2010). These studies emphasize the importance of taking N-C interactions into account when considering the possible impact of climate change on ecosystems, carbon sequestration, and in the development of earth system models (Thornton et al. 2009). Various mechanisms have been proposed to explain the changes in N-C interactions following increased N availability: biomass increase with allocation changes in plants (i.e., reduction of

below-ground C allocation due to the N increase) (Högberg et al. 2010); enhanced soil respiration reflecting an increase in soil microbial activity with an increase in N availability (Gärdenäs et al. 2011); inhibition of litter decomposition through a change in litter quality with elevated N (Knorr et al. 2005; Pregitzer et al. 2008); a change in plant uptake of organic N as a nutrient source in N-limited environments (Gärdenäs et al. 2011); altered interaction with dissolved organic carbon (e.g., enhanced mineralization of DOC due to increased abundance of electron acceptors in the form of NO₃⁻ in anoxic soil micro-sites (Kopáček et al. 2013c)); and changes in abiotic N-C interactions in soil (e.g., abiotic reaction of nitrite with dissolved organic matter through nitration and nitrosation of aromatic ring structures) (Davidson et al. 2003).

Phosphorus (P) is also an essential nutrient for biota. N-C-P interactions in soil vary among biomes.

Where P limits primary production, such as in some tropical ecosystems or acid alpine grasslands, increases in N deposition may have little impact on productivity (Matson et al. 1999), a finding that has recently been documented by field experiments at ILTER sites (Cusack et al. 2011; Bowman et al. 2008). In both N- and P- limited tundra ecosystems, C fluxes were found to respond positively to additions of both elements, although responses to P tended to be stronger than to N (Shaver et al. 1998). Bergström and Jansson (2006) have shown that increased N deposition may have changed lakes from N-limitation to P-limitation in remote and small lakes across the northern hemisphere, an observation supported by nutrient addition experiments in UK upland streams (Maberly et al., 2002). On the other hand, an assessment of long-term data from a Spanish ILTER lake site by Camarero and Catalan (2012) suggested that atmospheric P deposition may cause lakes to revert from P-limitation to N-limitation. There are clear needs for research into long-term C-N-P interactions for a much wider range of biomes.

Nitrogen deposition can serve as an acidifying as well as eutrophying agent (Oulehle et al. 2008).

Bowman et al. (2008) reported that long-term acid deposition in the Western Tatra Mountains of Slovakia, central Europe has altered soil systems in alpine grasslands to an extreme level of acidification usually

associated with soils exposed to acid mine drainage. They showed that increases in N deposition had resulted in a depletion of base cations, increases in aluminium (Al) and extractable iron (Fe) in soil, and a reduction in the biomass of vascular plants associated with a decrease in shoot calcium and magnesium concentrations. They suggested that acidifying soils in central Europe have reached an unprecedented level of toxicity in which Al release into soil water has been superseded by Fe release (Bowman et al. 2008).

N leaching in dormant season

Seasonal changes in nitrogen behavior of ecosystems are mostly driven by seasonal fluctuations of physical drivers (i.e. weather conditions) and biological factors (i.e., phenology in plant and microbial activity). The seasonality of plant growth in many biomes results in a seasonal N demand, while (with the exception of heavily N-saturated soils and catchments with little soil cover) most NO₃⁻ leached into surface and ground waters is increasingly being found to have undergone soil microbial processing (Stoddard 1994; Piatek et al. 2005; Curtis et al. 2012). Consequently, the dependence of biological systems on soil microclimate can lead to strong seasonal variation in N fluxes and concentrations in soils and drainage waters.

Time series analyses from boreal to temperate forested, moorland and alpine systems emphasize the importance of winter temperatures and snow cover in determining the export of NO₃⁻ in soil, ground and surface waters. Although winter has sometimes been considered to be the "dormant season," due to cold temperatures, vegetation dormancy and a persistent snow cover, microbial processes can persist and exert a critical impact on N cycling (Campbell et al. 2005; Makoto et al. 2013). Snow also allows solutes to accumulate in the soil (Kurian et al. 2012) leading to pronounced fluxes when the snow melts. Watersheds throughout the Northeast USA export more than 85% of the annual NO₃⁻ loss during winter

(Mitchell et al. 1996), with most of this export occurring during spring snowmelt (Campbell et al. 2005), but mid-winter melt events and rain-on-snow events can also influence winter NO₃⁻ loads to streams (Casson et al. 2010). Individual rain-on-snow events can contribute as much as 40% of annual NO₃⁻ export from forested watersheds in southeast Canada, and the contribution of rain-on-snow events to annual and winter NO₃⁻ loads has generally increased in recent decades (Eimers et al. 2007).

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The insulating properties of snow can maintain soil temperatures sufficiently high to allow root growth, microbial respiration and other biotic activities to continue (Groffman et al. 2009). Soils devoid of a snowpack are more vulnerable to freezing and hence to physical and chemical changes, including the death of fine roots, cell lysis and the alteration of soil microbial processes (Tierney et al. 2001; Christopher et al. 2008; Shibata et al. 2013). Experimental snow removal in alpine Europe lowered soil temperatures and increased NO₃ release (Freppaz et al. 2008). In mountainous and northern regions, soil temperatures and ecosystem respiration rates tend to be higher in winters with high amounts of snowfall than in winters with less or no snow (Monson et al. 2006). On the other hand, experimental snow manipulation at a mountain site in Norway indicated that no increase of inorganic N fluxes was associated with snow removal (Kaste et al. 2008), suggesting that the effects of decreased snow on the N cycle varies among locations. Coherent patterns of variation in NO₃ leaching are sometimes evident over large spatial scales and across catchments covering wide altitudinal gradients and land-use classes (e.g. Evans et al. 2010; Rogora et al. 2008). In the UK, winters when NO₃ leaching to remote surface waters have been strongly associated with negative excursions in the winter North Atlantic Oscillation (NAO) Index, most likely due to low winter temperatures, lower than average rainfall and larger contributions from more polluted air masses originating from the European continent (Monteith et al. 2000; George et al. 2004a). However, as soil temperature is a dominant driver, opposing inter-annual patterns in NO₃ leaching may be observed in regions normally blanketed by snow in winter when they lack snow cover, since snow cover insulates the uppermost soil layer from the atmosphere (Groffman et al. 2009; Makoto et al. 2013). Consequently, relationships between NO₃⁻ leaching and the NAO index in the UK and in northern Europe may vary regionally (George et al. 2004b; Blenckner et al. 2007; de Wit et al. 2008).

Over the last decade, trends in NO₃⁻ concentrations in waters across the UK Upland Waters Monitoring Network have lost coherence and begun to diverge (Monteith et al., 2014). Inter-annual variation in NO₃⁻ concentrations in some northern sites remains tightly linked to the winter NAO Index, and shows a long term (>20 years) increase consistent with the long term decline in the NAO over the same period (Fig. 3). At other sites further south, however, NO₃⁻ concentrations, while still showing evidence of influence of the NAO on short term variability, are trending downwards (Fig. 3), possibly in response to more marked reductions in N deposition in this region. The divergence provides evidence for regional differences in the relative importance of N deposition and climate variability on NO₃⁻ leaching to surface waters, with the latter exerting greater influence in areas where N deposition has been more stable.

Cascading influences from terrestrial watersheds to estuaries

Nitrogen losses from agricultural land are often several times higher than those from natural systems. In typical agrarian systems, exported N represents 10–40% (~25% on average) of net anthropogenic nitrogen inputs (Howarth et al. 2011), depending on the amounts of leachable NO₃⁻ in the soil and surplus water to transport the solutes out of the watershed. Kopáček et al. (2013a, b) indicated that, in Slapy Reservoir, an ILTER site in central Europe, the reservoir of leachable NO₃⁻ in agricultural and forested watersheds originates from both external (fertilization and atmospheric deposition) and internal (mineralization of soil organic N) sources, with relative contributions dependent on topography and land use practices such as drainage and tillage. Fluctuations in the export of N from Slapy Reservoir (Vltava river) from 1920–2010 were strongly related to the change in the mineralization of soil organic N

enhanced by more drainage of farmlands (up to 43%) rather than to the external N sources (Kopáček et al. (2013a, b). Boyer and Howarth (2002) evaluated the anthropogenic N source in large watersheds based on intensive monitoring of N cycles, indicating that fertilizer N inputs, N fixation in crop land and animal feed N imports were the dominant sources of Nr from agricultural land to the riverine N exports in sixteen large catchments of northeastern USA.

At watershed, landscape and regional scales, cascades and interactions are key to understanding N dynamics. Riparian zones function as the interface ("buffer zone") between terrestrial and aquatic ecosystems, and may act as either net sources or sinks of N depending on timescales, hydrological conditions and the history of N inputs. Anaerobic conditions in shallow groundwater in the riparian zone stimulate denitrification (microbial transformation of NO₃⁻ to reduced gaseous forms such as N₂O and N₂), thus reducing the potential flux of NO₃⁻ leaching to stream water (Chestnut and McDowell 2000). Hyporheic exchange, mixing saturated ground water (relatively anaerobic) with stream water (mostly aerobic), may provide "hot spots" for dynamic microbial N transformation near riparian boundaries and at the surfaces of channel beds (Shibata et al. 2004).

A substantial proportion of Nr can be buried in accumulating sediments of lakes and swamps (Noe and Hupp 2005). However, sediment anoxia may lead to the reduction of NO_3^- and nitrite (NO_2^-) to N_2 (or N_2O) by denitrification (Rissanen et al. 2011). Recently, "anaerobic ammonium oxidation" (anammox) has been identified as another process of N_2 release under anoxic conditions (Jetten et al. 2005).

In contrast to many freshwater systems, coastal systems may be N-limited as a consequence of eutrophication associated with high P inputs (Howarth and Marino 2006). Substantial denitrification rates have even been found in N-rich downstream riverine systems, whereas no comparable P-removal process occurs (Vermaat et al. 2012; Billen et al. 2009). Consequently, this may lead to N-limitation that enhances the dominance of diazotrophic cyanobacteria in periods when the high energy demand for N-fixation can be satisfied (Stal and Zehr 2008). This can even lead to temporal N limitation (Schubert et

al. 2010), whereby P-concentration increases during summer in response to the constant supply of P by runoff and release from the sediment (Figure 4; Schubert and Wasmund 2005). Further investigations of denitrification rates and their driving factors including regional comparisons of diverse riverine systems and their connections to marine systems would reduce the uncertainty of N budgets of coastal systems fed by terrestrial N inputs.

IMPACT ON BIODIVERSITY AND HUMAN SOCIETY

As described above, increases of anthropogenic Nr emission and deposition substantially alter N pools, cycles and transport among ecosystems. The altered N behavior also influences structures and characteristics of organisms in natural ecosystems and anthropogenically dominated systems. Here we review how increased Nr affects (i) biological structures in the context of biodiversity and also (ii) human society in the context of human health and ecosystem services.

Terrestrial biodiversity impacts

Numerous studies have reported a decline in species diversity (vascular plants, lichens, mosses, phytoplankton microbes, etc.) associated with both N fertilization and N gradients across a range of different ecosystem types in forest and semi-natural areas (Clark and Tilman 2008; Bobbink et al. 2010). Clark and Tilman (2008) demonstrated that chronic low-level nitrogen addition (10 kg N ha⁻¹ yr⁻¹ above ambient atmospheric N deposition) reduced plant species numbers by 17% relative to controls receiving ambient N deposition based on the multi-decadal experiment of N fertilization in the grassland ecosystem of Cedar Creek LTER. Recent studies suggested that even low-level N deposition could influence the changes in ground vegetation (Johansson et al. 2012; Hedwall et al. 2013). Nitrogen pollution stimulates

competitive interactions that lead to compositional change, making conditions unfavorable for some species (Bobbink et al. 2010; Bobbink and Hettelingh, 2011). Deprivation of light and nutrients resulting from the increased cover of aggressive dominant species can outweigh the potential benefits of N fertilization for subordinate species. These changes in biodiversity can have cascading impacts on primary production, soil C storage, microbial activity, rates of decomposition, N mineralization and immobilization, tissue chemistry, trophic interactions (herbivory), and can ultimately disrupt ecosystem functions and services (Ochoa-Hueso et al. 2011; Dias et al. 2011).

The sensitivity of terrestrial biodiversity to the deposition of oxidized and reduced N provides the basis for setting critical loads for N deposition both in Europe and North America (Cape et al. 2009; Pinho et al. 2011, 2012; Bobbink and Hettelingh, 2011). Independently-derived critical levels for lichens and moss diversity have been found to be similar for northern and southern Europe, thus emphasizing the universal applicability of these plant groups as ecological indicators of N deposition. Pardo et al. (2011) showed that, in the USA, empirical critical loads for N tend to increase according to the following sequence: lichens and bryophytes, mycorrhizal fungi, herbaceous plants and shrubs, and trees. In several studies in the USA, lichens invariably showed the lowest NH₃ critical levels (1 µg m⁻³) and N critical loads (1 kg N ha⁻¹ yr⁻¹) of all biological indicators (Jovan et al. 2012).

Currently, there is no coordinated global observation system capable of providing real-time tracking of biodiversity change around the globe (Pereira et al. 2013). The scientific understanding of how biodiversity is reacting to increasing N inputs, and how this is affecting ecosystem resilience and ecosystem services remains limited. However, biodiversity seems to be a relatively sensitive metric for measuring the effects of N at the ecosystem level, i.e., loss of particular species from an ecosystem (Cape et al. 2009). Changes in biodiversity can also be used to help identify those species most sensitive to increased N. We expect that various assessments of biodiversity will exhibit differences in scalability, temporal sensitivity, feasibility, and relevance.

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Human health

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Human-induced changes in response to changes in the N cycle also have both negative and positive implications for human health. The most obvious health benefit of increased use of Nr is decreased hunger and malnutrition through the use of fertilizers, while N-related air pollutants are hazardous for humans. Atmospheric N pollution can affect human health by increasing respiratory problems especially those caused by smaller particulate matter (PM_{2.5}), since they have the ability to penetrate deeper into the respiratory tract. Approximately 40% of PMs are NH₄⁺ and NO₃. High levels of atmospheric NO_x lead to increases in tropospheric O₃ that strongly affect human respiratory function (von Mutius 2000). In addition, high concentrations of NO2 in urban air can lengthen and worsen common viral infections such as human rhinovirus, significantly elevating the risks to asthmatics and individuals with compromised immune systems (Spannhake et al. 2002). Nitrogen also affects human health via water pollution. Concentrations of NO₃ in drinking water exceeding 10 mg L⁻¹ put children at risk of methemoglobinemia ("blue-baby" syndrome; Gupta et al. 2000). Even nitrate levels below the WHO standard may stimulate the endogenous formation of N-nitrosamines (van Mannen et al. 1996), compounds strongly implicated in cancer risks. Long-term consumption of water with NO₃ concentrations of 6.3 mg L⁻¹ and above has been linked to a higher risk for Non-Hodgkin's lymphoma (Ward et al. 1996). In Iowa, rising NO₃ levels well below the 10 mg L⁻¹ standard were associated with an increased risk of bladder and ovarian cancers (Weyer et al. 2001). One way to maintain good nourishment of the human population while decreasing fertilizer consumption would be to reduce meat consumption and increase consumption of a diversity of vegetables. Typically most NO₃ exposure (86%) to humans comes from vegetables, whereas the primary contributors to nitrite (NO₂) intake are cured meats (39%), baked goods and cereals (34%), and vegetables (16%). It is

possible that a diet based on a diversity of vegetables provides man with adequate levels of NO₂ which contributes to the whole-body NO production and homeostasis (Landberg et al. 2011).

Global food and feed trades are one of the important drivers of global, regional and local N circulation (Galloway et al. 2008). International trade in N has increased eightfold (from 3 to 24 Tg N) during 1961–2010 and a small number of countries (e.g., USA, Argentina and Brazil) currently feed much of the rest of the world in terms of proteins and N (Lassaletta et al. 2014). The spatial imbalance of production and consumption of feed and food contributes to the spatial imbalance of risk for human health and environment quality (e.g., animal feed imports from Africa, where the export of N contributed to N limitation of human food production with negative consequences for human health locally). Also, the influence of the global trade in N is more complex than merely the N flows associated with import or export of food and feed because part of the reactive N added by fertilizers and feed during the production of crop and animal products is lost to the environment, becoming a source of water and air pollution (Leach et al. 2012).

Impact on societal and economic value

The concept of ecosystem services (Millennium Ecosystem Assessment Board 2005) recognizes the vital importance of the natural environment and the biodiversity it supports in underpinning human wellbeing. The nitrogen cycle is central to several key ecosystem processes including: water quality regulation (regulating services); ecosystem productivity that is often limited by N (provision of services via food, timber, and fiber); C sequestration and control of N₂O emissions (via climate regulation services). Nitrogen also indirectly impacts all ecosystem services through its influence on biodiversity (e.g. Suddick and Davidson 2012). Clearly, a range of N-related ecosystem services may be threatened by anthropogenic disturbances of biogeochemical cycles.

To date, economic valuation of N-related ecosystem services and human health has been conducted mainly in Europe and the USA (Compton et al. 2011; Sutton et al. 2011), while assessment in other region such as Asia-Pacific, Africa and South America is still limited and entails great uncertainty. The European Nitrogen Assessment (ENA) estimated that the highest social costs of N are associated with air pollution effects of NOx on human health (10 to 30€ per kg of N). A similar value (\$28 per kg NOx-N) relating to the USA was obtained by Compton et al. (2011). The effects of N loss to water on aquatic ecosystems were evaluated by the ENA as: 5 to 20€ per kg of N. The ENA also estimated N-related environmental damage from agriculture in the EU to be 20 to 150 billion € year⁻¹, which is comparable with a benefit of N-fertilizer for farmers of 10 to 100 billion € per year⁻¹ (Sutton et al. 2011).

THE NEED FOR INTERNATIONAL INTEGRATION OF LONG-TERM ECOSYSTEM RESEARCH

Based on the above literature review, we propose that several areas require more attention to develop a better understanding and reduce uncertainties with respect to the environmental effects of N.

The long-term monitoring and analysis of N deposition, N cycles in various ecosystems, biodiversity, and N export to water are needed to provide the fundamental information necessary to address a spectrum of research questions concerning N dynamics in coupled human and ecological systems (Robertson et al. 2012; Driscoll et al. 2012). Modeling and analysis coupled with long-term monitoring and field experiments provide powerful research tools to help understand the dynamic features of the N cycle driven by multiple environmental factors, and to address new parameters to be monitored or examined (e.g., Aber et al. 2002; Driscoll et al. 2003). There are clearly regional research gaps in the context of long-term site-based research on N biogeochemistry: East Asia, South America and Africa (e.g. Anderson et al. 2012; Urakawa et al. 2012) where increased Nr pollution has been predicted for the

coming decades (Galloway et al. 2004). Also, long-term research sites in agricultural and urban ecosystems are currently limited despite their large importance in global N cycles. Increased international collaboration and integration offers the potential for further significant scientific advances, particularly with respect to the elucidation of: (i) responses of N_2O emission to elevated Nr inputs; (ii) biodiversity changes associated with changes in N deposition; (iii) spatial heterogeneity of temporal trends among different deposited N species; (iv) spatial patterns in N leaching from a wide spectrum of catchments including a range of altitudinal and latitudinal gradients and different land-use types; (v) long-term trends in N concentrations in surface waters and potential linkages with inter-annual climate change.

- Second, studying N dynamics at the ecosystem level should be coupled with socio-economic issues. Reactive nitrogen in the environment presents society with a global problem that must be addressed at a global scale over the long-term by uniting the analyses of natural and human systems. Continued maintenance of our best long-term environmental observation systems and the development of new long-term experiments will be necessary to clarify these complex interactions and their long- and short-term impacts.
- At the international scale, our environmental observation capacity remains extremely limited. Based on the analysis in this review, some research questions have emerged:
- To what extent do ecosystems exhibit common or unique responses to elevated Nr across different environmental and social landscapes?
- What features of socio-ecological N interactions are likely to be most sensitive to global changes in
 human population and climate?
- What are the political and management options to mitigate or adapt the N-related social issues (e.g. diet, human health and ecosystem services)?
- Will future climate change have major impacts on N biogeochemistry, and what feedbacks from N
 cycling will be most important in influencing the climate?

Even though the answers to these questions remain unclear, some strategies need to be developed. For example, some previous studies have suggested that very sensitive organisms, such as lichens and mosses could be effective early warning indicators of atmospheric Nr pollution in the early stages of anthropogenic disturbance of N cycles in an ecosystem (Pinho et al. 2011). The functional diversity of lichens and/or mosses coupled with measures of their nitrogen content and isotopic composition have the potential to be explicit spatial indicators of the early effects of Nr pollution., It would therefore be possible to use lichens and mosses in the long-term ecological site to develop an early-warning biological monitoring system of atmospheric N pollution in regional and global scales. Other responses of ecosystem structure and functioning to altered N cycle often vary among sites, influenced not only by current driving factors but also by long-term socio-ecological legacies (Aber et al. 1998). Therefore, investigations need to include international comparisons of the impact of socio-ecological legacies on current N cycling processes by analyzing the relationship between current monitoring data and previous land history and other parameters. The historical records of site-management, land-use, natural disturbances, climate, atmospheric deposition, etc. should be used to help to understand how historical factors are affecting current N cycles. Such analyses should be facilitated by using the results from long-term ecological sites such as the ILTER worldwide network. Meta-analysis of comprehensive and integrated N databases and the organization of workshops on focused topics using international researchers networks (e.g. ILTER) should be encouraged. The outcomes of collaborative international research should also include analyses of the complete spectrum of socio-ecological factors related to N. This information must to be provided to both the scientific community as well as other stakeholders, including policy makers.

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Ecosystems

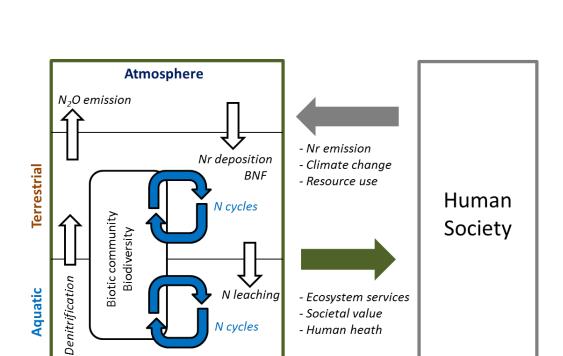


Fig.1 Conceptual framework of Nitrogen (N) biogeochemistry in coupled human and ecological systems in this review. Black arrows indicate anthropogenic disturbance and ecosystem feedbacks among both systems. White arrows represent dominant reactive nitrogen (Nr) flow; Nr deposition, Biological N fixation (BNF), N leaching and emission of nitrous oxide (N2O). Blue arrow shows N cycles among plant-soil-microbe systems.

- Human heath

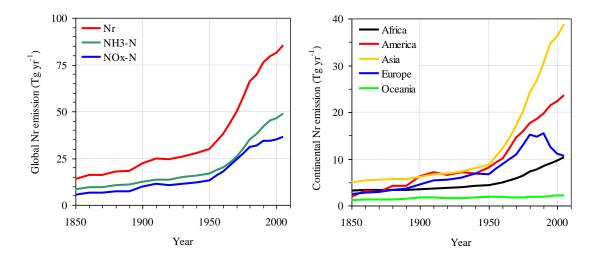


Fig.2 Global (left panel) and continental (right panel) annual rates of Nr (NO_x -N + NH_3 -N) emissions (derived from Kopáček and Posch, 2011).

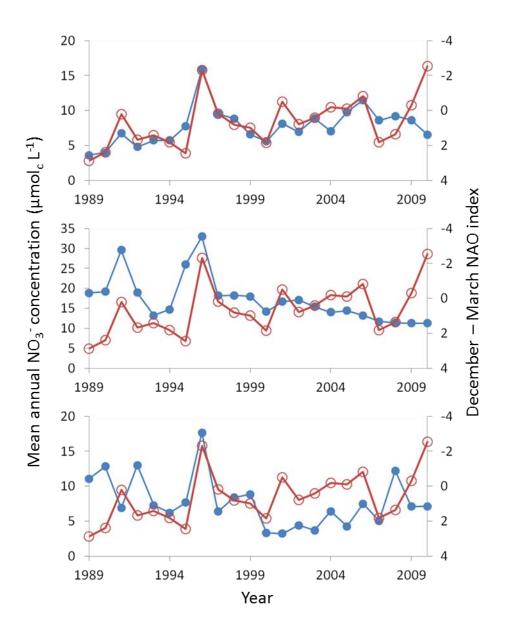


Fig. 3 Long-term observation of nitrate concentrations in 3 small upland UK lakes (The UK Upland Waters Monitoring Network).

From the top, the sites are Round Loch of Glenhead (southwest Scotland), Scoat Tarn (English Lake District), and Llyn Llagi (North Wales - Snowdonia). The blue circles represent the annual means of seasonal (4 samples per year) nitrate concentrations. The open red circles represent the December to March North Atlantic Oscillation Index. The NAO scale is reversed so the most negative values are uppermost.

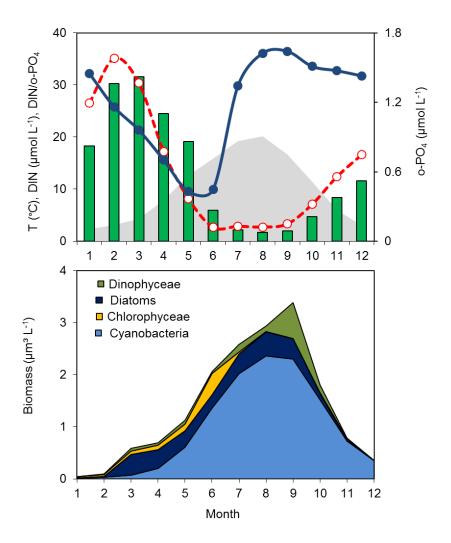


Fig. 4 Seasonality of dissolved nutrient concentration, temperature and phytoplankton biomass in a coastal water body. Shown are seasonal changes in dissolved inorganic nitrogen (DIN in μmol L⁻¹; open circles), ortho-phosphate (o-PO₄ in μmol L⁻¹; closed circles), temperature (T in °C; grey background area), DIN/o-PO₄ ratio (bars; upper panel) and phytoplankton biomass (μm³ L⁻¹, lower panel, resolved for main taxonomic groups) of a coastal inlet (Kubitzer Bodden and Strelasund, Southern Baltic Sea coast, Germany). All values are averages over a 10-year period (1988-1999). For further details see Schubert and Wasmund (2005).